

# Limiting Reagent Worksheet

## Answers

- ① Given the following reaction:



Balance the reaction.

- ① Suppose 14.8 g of  $\text{C}_3\text{H}_8$  reacts with 3.44 g of  $\text{O}_2$ . Determine the limiting reagent.

$$14.8 \text{ g C}_3\text{H}_8 \times \frac{1 \text{ mol C}_3\text{H}_8}{44.09 \text{ g C}_3\text{H}_8} \times \frac{4 \text{ mol H}_2\text{O}}{1 \text{ mol C}_3\text{H}_8} \times \frac{18.02 \text{ g H}_2\text{O}}{1 \text{ mol H}_2\text{O}} = 24.2 \text{ g H}_2\text{O}$$

$$3.44 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{32 \text{ g O}_2} \times \frac{4 \text{ mol H}_2\text{O}}{5 \text{ mol O}_2} \times \frac{18.02 \text{ g H}_2\text{O}}{1 \text{ mol H}_2\text{O}} = 1.55 \text{ g H}_2\text{O}$$

Since  $\text{O}_2$  produces less quantity of  $\text{H}_2\text{O}$  than  $\text{C}_3\text{H}_8$  does, it is the limiting reagent.

- ② Determine the amount of  $\text{CO}_2$  produced.

$$3.44 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{32 \text{ g O}_2} \times \frac{3 \text{ mol CO}_2}{5 \text{ mol O}_2} \times \frac{44.01 \text{ g CO}_2}{1 \text{ mol H}_2\text{O}} = 2.84 \text{ g CO}_2$$

- ② Given the following reaction:



Balance the reaction.

- ① What is the limiting reagent if 2.2 g of Mg reacts with 4.5 L of  $\text{O}_2$  at STP?

$$2.2 \text{ g Mg} \times \frac{1 \text{ mol Mg}}{24.3 \text{ g Mg}} \times \frac{2 \text{ mol MgO}}{2 \text{ mol Mg}} = 0.0905 \text{ mol MgO}$$

$$4.5 \text{ L O}_2 \times \frac{1 \text{ mol O}_2}{22.4 \text{ L O}_2} \times \frac{2 \text{ mol MgO}}{1 \text{ mol O}_2} = 0.401 \text{ mol MgO}$$

Since Mg produces less MgO than  $\text{O}_2$ , Mg is the limiting reagent.

- ② Determine the amount of MgO produced in grams.

$$0.095 \text{ mol MgO} \times \frac{40.31 \text{ g MgO}}{1 \text{ mol MgO}} = 3.83 \text{ g MgO}$$

# Limiting Reagent Worksheet 1f8766

**DJ Losen**



## **Limiting Reagent Worksheet 1f8766:**

The book delves into Limiting Reagent Worksheet 1f8766. Limiting Reagent Worksheet 1f8766 is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Limiting Reagent Worksheet 1f8766, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Limiting Reagent Worksheet 1f8766
- Chapter 2: Essential Elements of Limiting Reagent Worksheet 1f8766
- Chapter 3: Limiting Reagent Worksheet 1f8766 in Everyday Life
- Chapter 4: Limiting Reagent Worksheet 1f8766 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Limiting Reagent Worksheet 1f8766. The first chapter will explore what Limiting Reagent Worksheet 1f8766 is, why Limiting Reagent Worksheet 1f8766 is vital, and how to effectively learn about Limiting Reagent Worksheet 1f8766.
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4. In chapter 3, this book will examine the practical applications of Limiting Reagent Worksheet 1f8766 in daily life. The third chapter will showcase real-world examples of how Limiting Reagent Worksheet 1f8766 can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Limiting Reagent Worksheet 1f8766 in specific contexts. The fourth chapter will explore how Limiting Reagent Worksheet 1f8766 is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Limiting Reagent Worksheet 1f8766. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Limiting Reagent Worksheet 1f8766.

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