

# Limiting Reagent Worksheet

- 1 Nitric oxide (NO) reacts with oxygen gas (O<sub>2</sub>) to form nitrogen dioxide (NO<sub>2</sub>) gas.



In one experiment, 0.866 mol of NO is mixed with 0.503 mol of O<sub>2</sub>.

- Determine the limiting reagent.
- Calculate the number of moles of NO<sub>2</sub> produced.

- 2 Consider the following reaction:



- Suppose 0.86 mol of MnO<sub>2</sub> reacts with 48.2 g of HCl. Which reagent will be used first?
- How many grams of Cl<sub>2</sub> will be produced?

# Limiting Reagent Question And Answer Key

**F Rizvi**



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