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web startup write the neutralization reaction reactants write bottom the reactants one acid furthermore one base in any order HCl NaOH or furthermore NaOH HCl is fine

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web jan 22 2022 neutralization process you have learned that an acid is a compound that produces hydronium ions  $\text{H}_3\text{O}^+$   
and that a base is a compound that produces hydroxide ions  $\text{OH}^-$  strong acids and bases