

১৫। $x^2 - \sqrt{5}x + 1 = 0$ হলে $x + \frac{1}{x} =$ কত?

- (ক) $-2\sqrt{5}$ (খ) $-\sqrt{5}$
(গ) $\sqrt{5}$ (ঘ) $5\sqrt{5}$

১৬। $x + y = 6$ এবং $x - y = 4$ হলে $5xy$ এর মান কত?

- (ক) 16 (খ) 24
(গ) 25 (ঘ) 36

১৭। $U = \{1, 2, 3, 4, 5, 6, 7\}$, $A = \{3, 5, 6, 7\}$ হলে A র সমান কত হবে?

- (ক) $\{1, 2, 4\}$ (খ) $\{2, 4, 5\}$
(গ) $\{2, 4, 6\}$ (ঘ) $\{3, 5, 6, 7\}$

১৮। $A = \{3, 4, 5\}$ এবং $B = \{4, 5, 7\}$ হলে—

- i. $A \cup B = \{3, 4, 5, 7\}$
ii. $A \cap B = \{4, 5\}$
iii. $A \subset B$

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) i ও iii
(গ) ii ও iii (ঘ) i, ii ও iii

১৯। $x^2 - x - 42$ এর উৎপাদকে বিশ্লেষিত রূপ নিচের কোনটি?

- (ক) $(x-6)(x-7)$
(খ) $(x-6)(x+7)$
(গ) $(x+6)(x+7)$
(ঘ) $(x-7)(x+6)$

২০। কোনো বিন্দুটি x -অক্ষের উপর অবস্থিত?

- (ক) $(0, -4)$ (খ) $(-4, 0)$
(গ) $(0, 3)$ (ঘ) $(3, -1)$

২১। কোন প্রকৃত ভগ্নাংশের লব ও হরের যোগফল 14 এবং বিয়োজফল 8 হলে, ভগ্নাংশটি নিচের কোনটি?

- (ক) $\frac{5}{9}$ (খ) $\frac{9}{5}$
(গ) $\frac{3}{11}$ (ঘ) $\frac{11}{3}$

২২। $m^2 - 6m - 1 = 0$ হলে $m^2 + \frac{1}{m^2} =$ কত?

- (ক) 32
(খ) 34
(গ) 36
(ঘ) 38

২৩। একটি সমকোণী ত্রিভুজের বাহুদ্বয়ের ব্যাসার্ধ 4 সে.মি. এবং উচ্চতা 6 সে.মি.। ত্রিভুজটির বহু পৃষ্ঠের ক্ষেত্রফল কত? ($\pi = 3.14$)

- (ক) 75.36 বর্গ সে.মি.
(খ) 150.72 বর্গ সে.মি.
(গ) 226.08 বর্গ সে.মি.
(ঘ) 301.44 বর্গ সে.মি.

নিচের চিত্র অনুসারে ২৪ ও ২৫নং প্রশ্নের উত্তর দাও।



ABCD বর্জাক্ষেত্রের কর্ণদ্বয় 'O' বিন্দুতে মিলিত হয়েছে এবং $BC = 6$ cm.

২৪। $\angle AOB =$ কত?

- (ক) 30° (খ) 45°
(গ) 60° (ঘ) 90°

২৫। AC এর দৈর্ঘ্য কত?

- (ক) 6 সে.মি. (খ) $6\sqrt{2}$ সে.মি.
(গ) $9\sqrt{2}$ সে.মি. (ঘ) 12 সে.মি.

২৬। যে চতুর্ভুজের দুই জোড়া সম্মিলিত বাহু সমান তাকে কী বলে?

- (ক) অয়ত (খ) বর্গ
(গ) ঘূর্ণি (ঘ) সামান্তরিক

২৭।



ABC ত্রিভুজের ক্ষেত্রফল কত?

- (ক) 24 বর্গ সে.মি. (খ) 36 বর্গ সে.মি.
(গ) 48 বর্গ সে.মি. (ঘ) 60 বর্গ সে.মি.

২৮। একটি ত্রিভুজের বাহুদ্বয়ের অনুপাত $x : x : x\sqrt{2}$ হলে, এর বৃহত্তম কোণটির মান কত?

- (ক) 80° (খ) 85°
(গ) 90° (ঘ) 120°

২৯। চতুর্ভুজের চারটি কোণের সমষ্টি কত?

- (ক) 360° (খ) 300°
(গ) 270° (ঘ) 180°

৩০। একটি ট্রাপিজিয়ামের সমান্তরাল বাহু দুইটির দৈর্ঘ্য যথাক্রমে 13 সে.মি. এবং 8 সে.মি.। সমান্তরাল বাহুদ্বয়ের লম্ব দূরত্ব 4 সে.মি. হলে ট্রাপিজিয়ামটির ক্ষেত্রফল কত বর্গ?

- (ক) 42 বর্গ সে.মি. (খ) 52 বর্গ সে.মি.
(গ) 84 বর্গ সে.মি. (ঘ) 104 বর্গ সে.মি.

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75 and higher the coordination of trimethylamine by copper ii

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

stoichiometry lab copper sulfate iron pdf stoichiometry - Jun 01 2022

web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

stoichiometry lab the reaction of iron with copper ii sulfate - Oct 05 2022

web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

stoichiometry lab iron and copper ii sulfate ppt - Feb 26 2022

web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

stoichiometry the reaction of iron with copper ii sulfate - Jun 13 2023

web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

lab stoichiometry iron with copper ii sulfate pdf scribd - Nov 06 2022

web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

lab report 4 iron copper ii sulfate studocu - Aug 15 2023

web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

lab 4 lab 4 stoichiometry of iron copper ii sulfate redox - Jan 08 2023

web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

lab report 4 stoichiometry of iron copper ii sulfate - Dec 07 2022

web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

lab report 4 stoichiometry of iron copper ii sulfate - Feb 09 2023

web stoichiometric principles can be used to determine the chemical equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction the iron ions dissolve in the copper ii sulfate solution to produce copper ii ions

stoichiometry reaction of iron with copper ii sulfate - Apr 11 2023

web lab 4 stoichiometry the reaction of iron with copper ii sulfate revised 8 19 2009 1 introduction in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces

lab report copper iron stoichiometry 1522 words cram lab report - Mar 30 2022

web under the finalise of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

lab report stoichiometry the reaction of iron with copper ii sulfate - Aug 03 2022

web stoichiometry lab report the reaction of iron with copper ii sulfate period 5 in this experiment i observed the reaction of iron with copper ii sulfate i obtained the results to this experiment through adding heated CuSO_4 solution to the iron powder stirring the solution adding water then removing water adding acetone heating the acetone on low

stoichiometry the reaction of iron with copper ii sulfate - Jul 14 2023

web stoichiometry the reaction of iron with copper ii sulfate purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction

stoichiometry lab iron and copper sulfate studylib net - May 12 2023

web stoichiometry lab iron and copper sulfate sample data data and observations mass of copper ii sulfate pentahydrate 3 27 g of g mass of filter paper mass of filter paper and dry copper mass of dry copper 1 28 g

