6.03 Honors Chemical Reactions and Energy-Assignment Template

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Directions:

For this assessment, you will practice using potential energy diagrams. Respond to the questions below on energy diagrams. Then submit the answers to your teacher.

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Questions (40 points possible)

Consider the potential energy diagram shown below. This graph shows the chemical potential energy
in a reaction system over time. The y-axis is potential energy in kilojoules. The x-axis is the reaction
progress, or time.



A) Does this graph represent an endothermic or exothermic reaction? Explain your answer. (2 points)

Endothermic. The products are at a higher energy level than the reactants.

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Xiaolong Qi

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