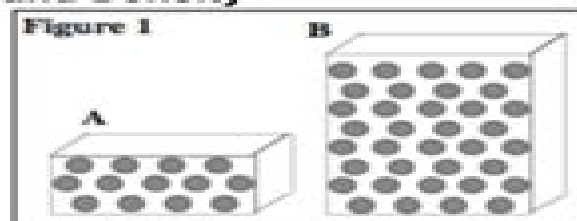


Chemistry – Unit 1 Worksheet 3

Mass, Volume, and Density

1. Study the matter shown in Figure 1. Each dot represents a particle of matter. [Assume the particles are uniformly distributed throughout each object, and particles of the same size have the same mass.]

- In the table below, show how the masses, volumes, and densities of A and B compare by adding the symbol $<$, $>$, or $=$ to the statement in the second column.
- Explain your reasoning for each answer in the last column.

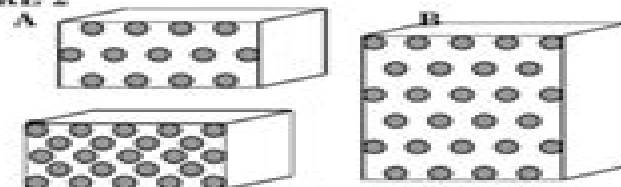


Property	Relationship	Reasoning
Mass	A $<$ B	There are fewer particles in A than in B.
Volume	A $<$ B	Object A occupies less space than B
Density	A $=$ B	The distribution of particles in both objects is equal.

2. Study the matter in Figure 2. [Assume the particles are uniformly distributed throughout each object, and particles of the same size have the same mass.]

- In the table below show how the masses, volumes, and densities compare by adding the symbol $<$, $>$, or $=$ to the statement in the second column.
- Explain your reasoning for each.

FIGURE 2



Property	Relationship	Reasoning
Mass	A $<$ B A $=$ C	There are fewer particles in A than in either B or C.
Volume	A $<$ B A $=$ C	Objects A & C occupy the same amount of space; object B occupies more space than A or C.
Density	A $=$ B A $<$ C	The distribution of the particles is the same in A & B; in C the particles are more tightly packed.

Modeling Chemistry Unit Ws 2

Brendan G. Carr



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