

Unit 6 – Worksheet 3

Ionic Compounds

Properties

Basic structural unit

1. Give the name of the following simple binary ionic compounds.

- a. Na_2O
 - b. K_2S
 - c. MgCl_2
 - d. CaBr_2
 - e. BaI_2
 - f. Al_2S_3
 - g. CsBr
 - h. AgF
2. Give the name of the following simple binary ionic compounds.
- a. Na_3N
 - b. K_2O
 - c. AgBr
 - d. MgI_2
 - e. SrO

3. Write the formula for the following binary ionic compounds.

- a. lithium bromide
- b. sodium iodide
- c. silver sulfide
- d. cesium oxide
- d. beryllium iodide
- f. barium hydride
- g. aluminum fluoride
- h. potassium oxide

4. Write the formula for these ionic substances.

- a. silver oxide
- b. aluminum sulfide
- c. sodium nitride
- d. barium chloride
- e. strontium hydride

Modeling Chemistry U6 Test A V 1

ML Morrison



Modeling Chemistry U6 Test A V 1:

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This book delves into Modeling Chemistry U6 Test A V 1. Modeling Chemistry U6 Test A V 1 is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Modeling Chemistry U6 Test A V 1, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Modeling Chemistry U6 Test A V 1
- Chapter 2: Essential Elements of Modeling Chemistry U6 Test A V 1
- Chapter 3: Modeling Chemistry U6 Test A V 1 in Everyday Life
- Chapter 4: Modeling Chemistry U6 Test A V 1 in Specific Contexts
- Chapter 5: Conclusion

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faire de la pâtisserie 3 boutique où l on fabrique vend et parfois où l on consomme des gâteaux 4 industrie activité

commerciale ayant trait à la fabrication et à la vente des gâteaux 5