

Chemistry – Unit 6 Worksheet 1

After having observed molecular models which show 18.098 compounds are discussed, the student should be asked to name positively charged ions, both named, write the molecular formulae, and to name negatively charged ions (anions). Then, students, using chemical names must write molecular formulae, and substitute the names "polyatomic ions" for such molecular effects to be tested. After each name, students are asked to write the chemical formula for each chemical element, and to write the elements that change off their table of elements.

1. Write the formula and name of the compound that is in the compound. The name and formula are given in parentheses.

Molecular formulae	1. sodium 2. oxygen	2. lithium 3. oxygen	2. aluminum 3. sulfur	2. calcium 3. sulfur
	<chem>Na2O</chem>	<chem>Li2O</chem>	<chem>Al2S3</chem>	<chem>CaS</chem>

name
formula



Molecular formulae	2. sodium 3. oxygen	1. magnesium 3. oxygen	2. calcium 3. sulfur
	<chem>Na2O</chem>	<chem>MgO</chem>	<chem>CaS</chem>

name
formula



Molecular formulae	1. magnesium 2. chlorine	2. sodium 3. fluorine	1. magnesium 2. bromine	1. sodium 2. chlorine
	<chem>MgCl2</chem>	<chem>NaF</chem>	<chem>MgBr2</chem>	<chem>NaCl</chem>

name
formula



Modeling Chemistry U6 Ws1 V2 0 Answers

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