MATTER—SUBSTANCES VS. MIXTURES

Tiron of Stollar

Name KEY

All matter can be classified as either a substance (element or compound) or a mixture (heterogeneous or homogeneous).

	Matte	r		
Substance can write chemical formula, homogeneous			Mixtures variable ratio	
Element one type atom	Compound two or more different atoms chemically bonded	Homogeneous solutions	Heterogeneous colloids and suspensions	

Classify each of the following as to whether it is a substance or a mixture. If it is a substance, write Element or Compound in the substance column. If it is a mixture, write Heterogeneous or Homogeneous in the mixture column.

	Type of Matter	Substance	Mixture
1.	chlorine	Element . Ma compound	
2.	water	Compound	
3.	soll		heterogeneou
4.	sugar water		homogeneous
5.	oxygen	Element or Compound	,
ó.	carbon dioxide	Element or Compound Compound	
7.	rocky road loe cre		haterogeness
8.	alcohol	Compound	
9.	pure air		homogeneous
10.	iron	Element.	/
Chem	histry IF8766	17	Sinstructional Fair, Inc.

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Robert C. Smoot, Smoot, Richard G. Smith, Jack Price

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