# Solubility Rules (General)

Soluble Ionic Compounds		Important Exceptions
Compounds containing	NO <sub>3</sub>	None
	C2H3O2	None
	CI	Compounds of Ag+, Hg22+, and Pb2+
	Br <sup>-</sup>	Compounds of Ag+, Hg22+, and Pb2+
	1-	Compounds of Ag <sup>+</sup> , Hg <sub>2</sub> <sup>2+</sup> , and Pb <sup>2+</sup> Compounds of Ag <sup>+</sup> , Hg <sub>2</sub> <sup>2+</sup> , and Pb <sup>2+</sup> Compounds of Ag <sup>+</sup> , Hg <sub>2</sub> <sup>2+</sup> , and Pb <sup>2+</sup>
	$SO_4^{2-}$	Compounds of Sr2+, Ba2+, Hg22+, and Pb2+
Insoluble Ionic Compounds		Important Exceptions
Compounds containing	S <sup>2-</sup>	Compounds of NH <sub>4</sub> <sup>+</sup> , the alkali metal cations, and Ca <sup>2+</sup> , Sr <sup>2+</sup> , and Ba <sup>2+</sup>
	CO <sub>3</sub> <sup>2-</sup>	Compounds of NH <sub>4</sub> <sup>+</sup> and the alkali metal cations
	PO <sub>4</sub> <sup>3-</sup>	Compounds of NH <sub>4</sub> <sup>+</sup> and the alkali metal cations
	OH-	Compounds of the alkali metal cations, and Ca <sup>2+</sup> , Sr <sup>2+</sup> , and Ba <sup>2+</sup>

# **Model I Rules Of Solubility In Aqueous Solutions**

SJ Ball

#### **Model I Rules Of Solubility In Aqueous Solutions:**

Ionic Surfactants and Aqueous Solutions Juan H. Vera, Grazyna Wilczek-Vera, 2018-07-09 Ionic Surfactants and Aqueous Solutions Biomolecules Metals and Nanoparticles covers a wide range of subjects related to aqueous systems from reverse micelles as ion exchangers to the study of micellar phase transfer catalysis for nucleophilic substitution reactions. The diverse background expertise and professional interests of the contributors to this book give to it a unique richness of approach in topics of relevance for biotechnology and environmental studies Over sixty publications presenting research results are combined and expanded in this book by some of the original researchers At a mature age and at the summit of successful professional careers they have taken a second look to the state of the art in the fields that they had pioneered Eva Rodil and Ana Soto who had their research formation in the group of Professor Alberto Arce at Universidade de Santiago de Compostela Spain are presently professors at that university Maen Husein is a professor at University of Calgary Canada Remy Dumortier Mohammad Khoshkbarchi Hamid Rabie and Younok Dumortier Shin are presently active leaders in the industrial world in Canada and the USA The editors are retired academics from McGill University Montreal Canada and coauthors of the book Classical Thermodynamics of Fluid Systems Industrial Waste Air Model (IWAIR) ,2002 Capture Processes Zhien Zhang, Tohid N. Borhani, Muftah H. El-Naas, Salman Masoudi Soltani, Yunfei Yan, 2020-12-14 This book introduces the recent technologies introduced for gases capture including CO2 CO SO2 H2S NOx and H2 Various processes and theories for gas capture and removal are presented The book provides a useful source of information for engineers and specialists as well as for undergraduate and postgraduate students in the fields of environmental and chemical **Industrial & Engineering Chemistry**, 1923 science and engineering **Industrial and Engineering Chemistry** Annual Reports on the Progress of Chemistry Chemical Society (Great Britain), 1908 ,1924 Introduction to Physical Chemistry Sir James Walker,1927 Theoretical Chemistry from the Standpoint of Avogadro's Rule and Thermodynamics Walther Nernst, 1904 Industrial Waste Air Model Technical Background Document ,2002 The Phase Rule and Its Applications Alexander Findlay, 1923 The Chemical News and Journal of Physical Science ,1911 Chemical News and Journal of Physical Science ,1895 British Chemical Abstracts ,1926 **British** Chemical News and Journal of Industrial Science Abstracts .1926 British Chemical and Physiological Abstracts, 1928 ,1887 Pharmaceutical Journal, 1908 Chemical Abstracts ,1908 Science Abstracts .1902 I & EC, 1925

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