# Solubility Rules for Common Ionic Compounds in Water Some Soluble Compounds Exceptions

Group IA cations (Li+, Na+, K+. Rb+, Cs+), NH4+

No Exceptions

No Exceptions

Halides of Ag+, Hg22+, Pb2+

Sulfates of Sr2+. Ba2+, Hg22+, Pb2+

Sulfides of Group 1A cations, NH4+, Ca2+,

Carbonates of Group 1A cations and NH<sub>4</sub>+

NO<sub>3</sub>. ClO<sub>4</sub>, CH<sub>3</sub>COO, ClO<sub>3</sub> Cl., Br : 1

Some Insoluble Compounds

SO42-

52-

POA3-

CO32-OH.

Hydroxides of Group 1A cations, NH<sub>4</sub>+, Ca2+, Sr2+, Ba2+

Sr2+, Ba2+

Phosphates of Group 1A cations and NH<sub>4</sub><sup>+</sup>

# **Model 1 Rules Of Solubility In Aqueous Solutions**

**M** Lipman

Model 1	Rules	Of So	lubility	In	Aqueous	<b>Solutions:</b>
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