Chemistry - Unit 6 Notes 1 - Thomson Model of the Atom

J. J. Thomson performed experiments with cathode rays in an attempt to understand electricity – which was still a mystery in the late 1800s. Use the website A Look Inside the Atom! to find the conclusions that Thomson and other physicists drew regarding the mysterious cathode rays.

<u>Thomson's 1897 Experiments</u> - state the conclusions Thomson drew from each of his famous cathode ray experiments:

- First Experiment: Thomson directed the beam at an electrometer and tried to separate
 the evidence of charge from the path of the beam. What connection did Thomson find
 between charge and the cathode rays? Was the charge positive or negative?
- 2. Second Experiment: Thomson tried passing the cathode ray through an electric field. How did cathode ray beam behave when it passed through an electric field? What did he conclude after his second experiment?
- 3. Third Experiment: Thomson did some careful measurements on how much the path of the cathode ray was bent in a magnetic field and how much energy they carried. From this work Thomson could describe the mass/charge ratio of the cathode ray particles. What amazing result did Thomson find?

Thomson's Atomic Model: Thomson presented three hypotheses from his experiments. Only two were accepted by physicists — in fact the third was shown to be wrong! From the first two came a model of the atom known as the *Plum Pudding* model. Complete the atom drawing below to illustrate Thomson's plum pudding model. Explain how this fits with his observations.



https://www.aip.org/history/exhibits/electron/jjrays.htm @Modeling Instruction - AMTA 2013

Modeling Chemistry U6 Review V1 Answers

Zhe Xu

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Whispering the Strategies of Language: An Mental Quest through Modeling Chemistry U6 Review V1 Answers

In a digitally-driven earth where displays reign supreme and immediate interaction drowns out the subtleties of language, the profound strategies and mental subtleties concealed within phrases often go unheard. However, nestled within the pages of **Modeling Chemistry U6 Review V1 Answers** a captivating literary prize sporting with natural emotions, lies an extraordinary journey waiting to be undertaken. Penned by a talented wordsmith, that charming opus invites readers on an introspective journey, delicately unraveling the veiled truths and profound affect resonating within the very fabric of each and every word. Within the psychological depths of the emotional review, we can embark upon a honest exploration of the book is core themes, dissect their interesting writing design, and yield to the powerful resonance it evokes heavy within the recesses of readers hearts.

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Table of Contents Modeling Chemistry U6 Review V1 Answers

- 1. Understanding the eBook Modeling Chemistry U6 Review V1 Answers
 - The Rise of Digital Reading Modeling Chemistry U6 Review V1 Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Modeling Chemistry U6 Review V1 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - $\circ\,$ Features to Look for in an Modeling Chemistry U6 Review V1 Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Modeling Chemistry U6 Review V1 Answers

- Personalized Recommendations
- Modeling Chemistry U6 Review V1 Answers User Reviews and Ratings
- Modeling Chemistry U6 Review V1 Answers and Bestseller Lists
- 5. Accessing Modeling Chemistry U6 Review V1 Answers Free and Paid eBooks
 - Modeling Chemistry U6 Review V1 Answers Public Domain eBooks
 - Modeling Chemistry U6 Review V1 Answers eBook Subscription Services
 - Modeling Chemistry U6 Review V1 Answers Budget-Friendly Options
- 6. Navigating Modeling Chemistry U6 Review V1 Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Modeling Chemistry U6 Review V1 Answers Compatibility with Devices
 - Modeling Chemistry U6 Review V1 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Modeling Chemistry U6 Review V1 Answers
 - Highlighting and Note-Taking Modeling Chemistry U6 Review V1 Answers
 - Interactive Elements Modeling Chemistry U6 Review V1 Answers
- 8. Staying Engaged with Modeling Chemistry U6 Review V1 Answers
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modeling Chemistry U6 Review V1 Answers
- 9. Balancing eBooks and Physical Books Modeling Chemistry U6 Review V1 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modeling Chemistry U6 Review V1 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Modeling Chemistry U6 Review V1 Answers
 - Setting Reading Goals Modeling Chemistry U6 Review V1 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modeling Chemistry U6 Review V1 Answers

- Fact-Checking eBook Content of Modeling Chemistry U6 Review V1 Answers
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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