

## JEE Advanced 2020 | Paper-2 | Code-E

4. An acidified solution of potassium chromate was layered with an equal volume of amyl alcohol. When it was shaken after the addition of 1 mL of 3%  $\text{H}_2\text{O}_2$ , a blue alcohol layer was obtained. The blue color is due to the formation of a chromium (VI) compound 'X'. What is the number of oxygen atoms bonded to chromium through only single bonds in a molecule of X?

Answer: 4

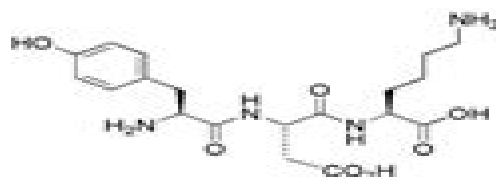
Solution:

Blue colour compound of 'Cr' is  $\text{CrO}_5$ .



Oxygen atoms bonded to chromium through only single bonds = 4

5. The structure of a peptide is given below.



If the absolute values of the net charge of the peptide at  $\text{pH} = 2$ ,  $\text{pH} = 6$ , and  $\text{pH} = 11$  are  $|z_1|$ ,  $|z_2|$ , and  $|z_3|$ , respectively, then what is  $|z_1| + |z_2| + |z_3|$ ?

Answer: 5

Solution:

(i) At  $\text{pH} = 2$  (Highly acidic)

# Jee Exam Paper 2 Code K Solution

**G Orfield**



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