

Question 9. How many oxygen atoms are there in 18 g of water?

Answer: Molar mass of water is 18 g/mol.

Number of oxygen atoms in 18 g of water = 6.02×10^{23}

Question 10. Name two factors that introduce uncertainty into measured figures.

Answer: (i) Reliability of measuring instrument.

(ii) Skill of the person making the measurement.

Question 11. State Avogadro's law.

Answer: Equal volumes of all gases under the conditions of same temperature and pressure contain the same number of molecules.

Question 12. How are 0.5 ml of NaOH different from 0.5 M of NaOH?

Answer: 0.5 ml of NaOH means 0.5 mole (20.0 g) of NaOH, 0.5M of NaOH means that 0.5 mole (20.0g) of NaOH are dissolved in 1L of its solution.

Question 13. What is one a.m.u. or one 'u'?

Answer: 1 a.m.u. or 1 u = $1/12$ th mass of an atom of carbon 12.

Question 14. What is the number of significant figures in 1.050×10^4 ?

Answer: Four.

II. Short Answer Type Questions

Question 1. Define molality. How does molality depend on temperature?

Answer: Molality is defined as the moles of solute per kilogram of solvent.

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Karen C Timberlake



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Student Guide For Living Chemistry Barbara Ucko, 2012-12-02 Student Guide for Living Chemistry is a 23 chapter textbook guide that allows students to study and review on their own and test their understanding to help them prepare for examinations Every chapter begins with a list of objectives stating exactly the skills to develop in a particular unit Each objective corresponds to a section in the textbook Living Chemistry Three kinds of questions are provided for each objective to check the student's understanding namely short answer Study Questions multiple choice and fill in The answers for all questions are provided at the end of the chapter The opening chapters cover the SI units composition of matter chemical bonding compounds chemical change gases respiration and water The subsequent chapters deal with solutions acids bases salts nuclear and organic chemistry oxygen derivatives and hydrocarbons polymers and other organic derivatives This textbook also explores the chemistry of carbohydrates lipids proteins enzymes and energy and carbohydrate metabolism The remaining chapters discuss the chemistry of vitamins hormones body fluid drugs and poisons Undergraduate chemistry students will find this book invaluable

Unplugging the Classroom Hilary Anne Wilder, Sharmila Pixy Ferris, 2017-05-17 Unplugging the Classroom Teaching with Technologies to Promote Students Lifelong Learning provides techniques to help teaching and learning in an age where technology untethers instruction from the classroom from semester seat time and from a single source of expertise The book brings together researchers and practitioners from diverse academic fields including library perspectives and presents interdisciplinary discussions from both theoretical and applied areas It is unique in its goal of bringing educators and librarians together to explore the challenges that are faced by students and faculty in any time any place any path and any pace learning In spite of the fact that the mobile revolution has definitively arrived students and faculty alike aren't ready to make the leap to mobile learning The pressures of technological advances along with the changing nature of learning will demand increasingly profound changes in education Researchers have begun to address this issue but the revolution in mobile communication has not been accompanied by a concomitant growth in pedagogical resources for educators and students More importantly such growth needs to be undergirded by sound learning theories and examples of best practice Provides a hands on resource useful to both novices and experts for technology enabled teaching and learning Gives both discipline specific and cross disciplinary perspectives Discusses discipline specific mobile applications Offers an opportunity to meet the needs of contemporary learners and foster their competencies as lifelong learners Addresses emerging issues in technology and pedagogy

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because US students are often poorly prepared when they reach the university level The chapter by Taagepera and Arasasingham deals with the application of knowledge spaces independent of ALEKS to the teaching of college chemistry The four chapters by Albert and his collaborators strive to give cognitive interpretations to the combinatoric structures obtained and used by the ALEKS system The contribution by Eppstein is technical and develops means of searching the knowledge structure efficiently *Organic Chemistry* T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-01-19 The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

Fundamentals of Chemistry Ralph A. Burns, 2003 For one semester preparatory chemistry courses or general purpose introductory chemistry courses This clearly written well illustrated versatile textbook provides thorough coverage of chemistry with a balance of problem solving skills real world applications and an emphasis on critical thinking and the process of science A supporting theme throughout the text continually emphasizes that chemistry is everywhere **Books in Print Supplement**, 2002 **The Best of the Teaching Professor** Magna Publications Incorporated, 2017-09-29

Excerpts from the newsletter The teaching professor **Children's Books in Print, 2007**, 2006 New York Times Saturday Book Review Supplement, 1966 **The Teaching of Chemistry** Norman Frederick Newbury, 1958 The Cumulative Book Index, 1996 A world list of books in the English language Video Computing, 1985 *Chemistry* Catherine E. Housecroft, Edwin C. Constable, 2010 This text integrates the three major branches of chemistry with the aim of enabling students to tackle more easily the problems within the subject and to apply chemistry to real life situations

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