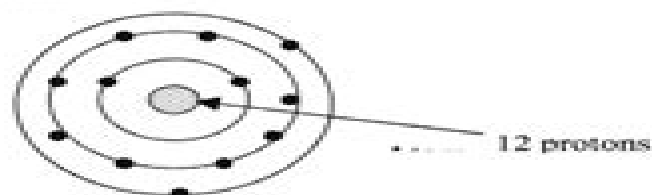


Forming Ions

Name: _____ Date: _____ Period: _____

Information: Ions

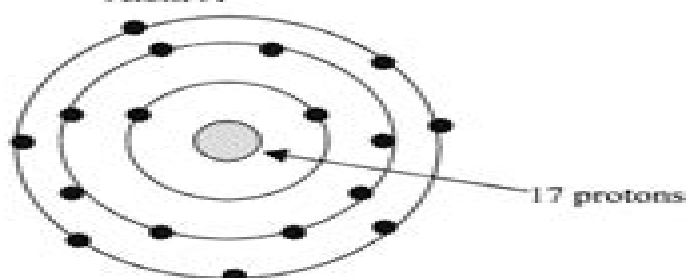
Figure 1: Below are four Bohr diagrams of atoms and ions. The two diagrams on the left are atoms; the two on the right are ions.



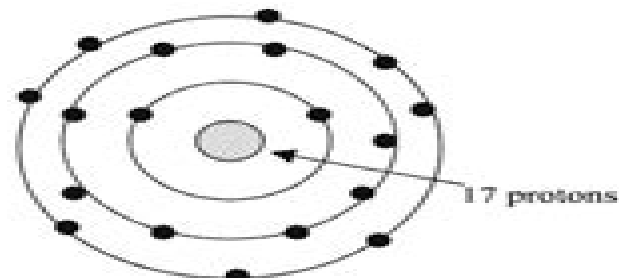
Atom A



Ion of Atom A



Atom B



Ion of Atom B

Critical Thinking Questions

1. How do you know that both Atom A and Atom B are neutral (have a charge of zero)?

2. What elements are Atom A and Atom B?

3. Given the above diagrams, explain how an atom becomes an ion?

4. What is the charge on Ion A? What is the charge on Ion B? How do you know?

5. Write the electron configuration (ex: $1s^2 2s^2 2p^6 \dots$) for each ion and atom shown in the Bohr diagrams.

Atom A: _____

Ion A: _____

Atom B: _____

Ion B: _____

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CL Gary

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'CL Gary' text.

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