

1. Use the VSEPR theory to predict molecular geometry of IF_5 .

a) tetrahedral	c) trigonal bipyramidal	e) octahedral
b) see-saw	d) square pyramidal	

a) bent c) trigonal pyramidal e) trigonal planar
b) tetrahedral d) T-shaped

a) 100 b) 120 c) 110 d) 90 e) 80

a) non-saw c) trigonal planar e) T-shaped
b) trigonal pyramidal d) tetrahedral

a) linear c) trigonal planar e) T shaped
b) bent d) tetrahedral

a) trigonal planar c) trigonal bipyramidal e) square planar
b) tetrahedral d) octahedral

a) trigonal planar c) trigonal bipyramidal e) square planar
b) tetrahedral d) octahedral

a) 90 e) none of these
b) 109.5 d) 180

For the following 4 questions, the following molecular substances are referred to. Any substance may be used more than once in answering the questions. CO_2 , SO_2 , SiF_4 , XeF_4 .

a) CO_2 c) SiF_4 e) none of these
 b) SO_2 d) XeF_4

a) CO_2 c) SiF_4 e) none of these
b) SO_2 d) XeF_4

a) CO_2 c) SF_6 e) none of these
b) SO_2 d) 2SF_6

☐ a. SO_2
☐ b. SO_3
☐ c. SF_6
☐ d. SOF_2

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JR Anderson



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