MCAT Chemistry - Medical College Admission Test Preparation



CHAPTER # 24

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- Benzene is soluble in
 - (a) alcohol
 - (c) ether

- (b) petrol
- (d) v all of these

- 2. Benzene is insoluble in
 - (a) V water

(b) alcohol

(c) petrol

(d) ether

- 3. Chlorination of benzene is:
 - (a) **V** An electrophilic substitution reaction (b)
 - (c) A free-radical reaction

- (b) A nucleophilic substitution reaction
- (d) An elimination reaction
- 4. The total number of sigma bonds in benzene are:
 - (a) 6

(b) √ 12

(c) 18

- (d) 3
- The Pi (π) molecular orbital in benzene contains:
 - (a) 2 electrons

(b) 4 electrons

(c) √ 6 electrons

- (d) 3 electrons
- 6. The C C C bond angles in benzene are:
 - (a) V 120°

(b) 90°

(c) 109.28°

- (d) 180°
- 7. The carbon-carbon bond length in benzene is:
 - (a) longer than a C C single bond
- (b) √ longer than a C = C double bond
- (c) Shorter than a C = C double bond
- (d) Shorter than a C 0 ≡ C triple bond
- 8. Two adjacent substitents on benzene ring is said to be:
 - (a) V Ortho to one another

(b) Meta to one another

(c) Para to one another

(d) None of the above

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Arshad Iqbal

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 - Chapter 2: Essential Elements of Mcgs In Aromatic Compounds
 - Chapter 3: Mcqs In Aromatic Compounds in Everyday Life
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- 3. In chapter 2, the author will delve into the foundational concepts of Mcqs In Aromatic Compounds. This chapter will elucidate the essential principles that need to be understood to grasp Mcqs In Aromatic Compounds in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Mcqs In Aromatic Compounds in daily life. The third chapter will showcase real-world examples of how Mcqs In Aromatic Compounds can be effectively utilized in everyday scenarios.
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 - This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Mcqs In Aromatic Compounds.

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