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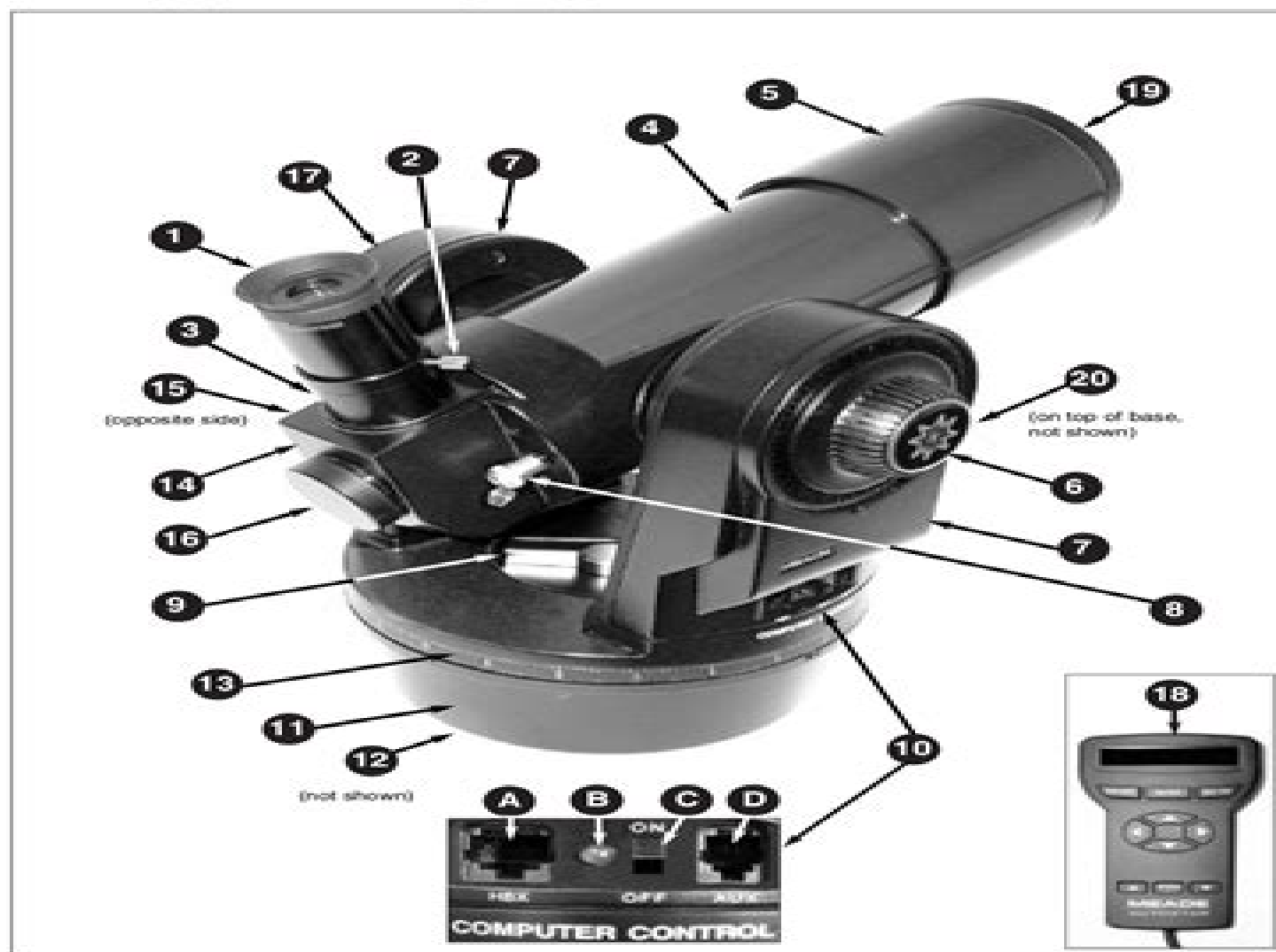


Fig. 1: The ETX-60AT and ETX-70AT Telescope.

Meade Autotrak Telescope Etx 60 Manual

Mike Weasner



Meade Autotrak Telescope Etx 60 Manual:

Using the Meade ETX Mike Weasner, 2002-01-25 The Meade ETX range of telescopes is one of the most successful ever made It is low cost has sold in its tens of thousands and is available in almost every country Here ETX expert Mike Weasner reveals everything any amateur astronomer ever wanted to know about the telescope First book dedicated entirely to the ETX Written by an acknowledged world authority Describes the best 100 objects to begin observing Contains detailed hints and tips aimed at getting the best out of the ETX Features imaging photographic and digital as well as visual observing

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web qualitative test flame tests introduction for this test i had five containers labelled a e each containing a metal salt i had to find out which metal is used in each of the five unknown salts by using a metal loop placing it in the unknown salts and testing it by putting the metal under a roaring bunsen flame safety points wear goggles all the time when

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web the easiest colors to observe during the flame test were that of copper chloride and lithium chloride for potassium chloride there was a purple color close to the stick but the flame remained orange for the majority of the burn time the unknown slat burned orange but also had a slight blue color close to the stick

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web use crayons or colored pencils to color in the spectrum in the box on your report sheet to show the correlation between color and wavelength part ii flame tests you will conduct flame tests to observe the flame emission colors for the following solutions lino3 cu no3 2 sr no3 2 ba no3 2 kno3 and nano3 since nitrates do not emit color you

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web task use a flame test to determine which ion salt produces the red color then using a spectroscope match the bright line spectra from red fireworks to a substance in your material list then make your final determination of what ion in the material list causes red fireworks procedure

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