

Sodium becomes a cation, and chlorine becomes an anion.

Activity A: Ions	Get the Gizmo ready: <ul style="list-style-type: none"> Click Reset. Check that sodium and chlorine are still selected. 	
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Introduction: Some of the particles that make up atoms have an electrical charge. Electrons are negatively charged, while protons are positively charged. Particles with opposite charges (+ and -) attract, while particles with the same charge (+ and + or - and -) repel.

Question: What happens when atoms gain or lose electrons?

1. **Count:** Electrons move around the nucleus of atoms in specific shells, shown by the rings around the atoms in the Gizmo. The first ring holds two electrons, and the second holds eight. (Electrons in the inner rings are not shown; you can assume these rings are full.)

- A. Observe the sodium and chlorine atoms. Assuming that the inner rings are full of electrons, how many electrons are there total in each atom?

Sodium: 11 electrons

Chlorine: 17 electrons

- B. Each atom is neutrally charged, which means that each atom has the same number of protons and electrons. Based on this, how many protons are in each atom?

Sodium: 11 protons

Chlorine: 17 protons

2. **Observe:** Most atoms are stable with a configuration of eight valence electrons. This is known as the **octet rule**. How many valence electrons does each atom have?

Sodium: 1 valence electron

Chlorine: 7 valence electrons

3. **Form a bond:** Each electron has a charge of $1-$, and each proton has a charge of $1+$. You can calculate the charge of an atom by subtracting the number of electrons from the number of protons. Move an electron from the sodium to the chlorine atom.

- A. What are the charges of each atom now? Sodium: +1 Chlorine: -1

Turn on **Show charge** to check. These charged atoms are called **ions**.

- B. Is each ion stable? Explain.

Yes, each ion is stable. Chlorine only has 7 valence electrons, so gaining 1 more would lead to a stable full octet. Sodium on the other hand has 1 valence electron, by losing this electron it has just a regular shell of 8 electrons, also becoming a stable octet.

Click **Check** in the lower right corner of the Gizmo to check.

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Ionic Bonds Gizmo Explore Learning Answers

Huangqi Zhang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Ionic Bonds Gizmo Explore Learning Answers

1. Understanding the eBook Ionic Bonds Gizmo Explore Learning Answers
 - The Rise of Digital Reading Ionic Bonds Gizmo Explore Learning Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Bonds Gizmo Explore Learning Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Bonds Gizmo Explore Learning Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Bonds Gizmo Explore Learning Answers
 - Personalized Recommendations
 - Ionic Bonds Gizmo Explore Learning Answers User Reviews and Ratings
 - Ionic Bonds Gizmo Explore Learning Answers and Bestseller Lists
5. Accessing Ionic Bonds Gizmo Explore Learning Answers Free and Paid eBooks
 - Ionic Bonds Gizmo Explore Learning Answers Public Domain eBooks
 - Ionic Bonds Gizmo Explore Learning Answers eBook Subscription Services
 - Ionic Bonds Gizmo Explore Learning Answers Budget-Friendly Options

6. Navigating Ionic Bonds Gizmo Explore Learning Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Bonds Gizmo Explore Learning Answers Compatibility with Devices
 - Ionic Bonds Gizmo Explore Learning Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Bonds Gizmo Explore Learning Answers
 - Highlighting and Note-Taking Ionic Bonds Gizmo Explore Learning Answers
 - Interactive Elements Ionic Bonds Gizmo Explore Learning Answers
8. Staying Engaged with Ionic Bonds Gizmo Explore Learning Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Bonds Gizmo Explore Learning Answers
9. Balancing eBooks and Physical Books Ionic Bonds Gizmo Explore Learning Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Bonds Gizmo Explore Learning Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ionic Bonds Gizmo Explore Learning Answers
 - Setting Reading Goals Ionic Bonds Gizmo Explore Learning Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ionic Bonds Gizmo Explore Learning Answers
 - Fact-Checking eBook Content of Ionic Bonds Gizmo Explore Learning Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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