

FORMULAS FOR IONIC COMPOUNDS

1. Write the chemical formula for the ionic compounds made from the following elements.

Elements	metal ion	non-metal ion	Chemical Formula	Name
potassium and iodine	K^{+1}	I^{-1}	KI	Potassium Iodide
barium and chlorine	Ba^{2+}	Cl^{-1}	$BaCl_2$	Barium Chloride
sodium and oxygen	Na^{+1}	O^{2-}	Na_2O	sodium Oxide
aluminum and nitrogen	Al^{3+}	N^{3-}	AlN	Aluminum Nitride
lithium and sulfur	Li^{+1}	S^{2-}	Li_2S	Lithium Sulfide
magnesium and oxygen	Mg^{+2}	O^{2-}	MgO	Magnesium Oxide
calcium and chlorine	Ca^{2+}	Cl^{-1}	$CaCl_2$	calcium chloride
boron and chlorine	B^{3+}	Cl^{-1}	BCl_3	boron chloride
beryllium and sulfur	Be^{2+}	S^{2-}	BeS	beryllium sulfide

2. Complete the chart below and write the ionic formulas to complete it. A few have been done for you.

	F	Cl	O^{2-}	S^{2-}	N^{3-}	P^{3-}
Li^{+}	LiF	LiCl	Li_2O	Li_2S	Li_3N	Li_3P
Na^{+}	NaF	NaCl	Na ₂ O	Na_2S	Na_3N	Na_3P
K^{+}	KF	KCl	K_2O	K_2S	K_3N	K_3P
Be^{2+}	BeF_2	$BeCl_2$	BeO	BeS	Be_3S_2	Be_3P_2
Mg^{2+}	MgF_2	$MgCl_2$	MgO	MgS	Mg_3N_2	Mg_3P_2
Ca^{2+}	CaF_2	$CaCl_2$	CaO	CaS	Ca_3N_2	Ca_3P_2
Al^{3+}	AlF_3	$AlCl_3$	Al_2O_3	Al_2S_3	AlN	AlP
Ga^{3+}	GaF_3	$GaCl_3$	Ga_2O_3	Ga_2S_3	GaN	GaP
B^{3+}	BF_3	BCl_3	B_2O_3	B_2S_3	BN	BP

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J Elliott



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Table of Contents Ionic Compound Reviewing Vocabulary Answers

1. Understanding the eBook Ionic Compound Reviewing Vocabulary Answers
 - The Rise of Digital Reading Ionic Compound Reviewing Vocabulary Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Compound Reviewing Vocabulary Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Compound Reviewing Vocabulary Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Compound Reviewing Vocabulary Answers
 - Personalized Recommendations
 - Ionic Compound Reviewing Vocabulary Answers User Reviews and Ratings

- Ionic Compound Reviewing Vocabulary Answers and Bestseller Lists
- 5. Accessing Ionic Compound Reviewing Vocabulary Answers Free and Paid eBooks
 - Ionic Compound Reviewing Vocabulary Answers Public Domain eBooks
 - Ionic Compound Reviewing Vocabulary Answers eBook Subscription Services
 - Ionic Compound Reviewing Vocabulary Answers Budget-Friendly Options
- 6. Navigating Ionic Compound Reviewing Vocabulary Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Compound Reviewing Vocabulary Answers Compatibility with Devices
 - Ionic Compound Reviewing Vocabulary Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Compound Reviewing Vocabulary Answers
 - Highlighting and Note-Taking Ionic Compound Reviewing Vocabulary Answers
 - Interactive Elements Ionic Compound Reviewing Vocabulary Answers
- 8. Staying Engaged with Ionic Compound Reviewing Vocabulary Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Compound Reviewing Vocabulary Answers
- 9. Balancing eBooks and Physical Books Ionic Compound Reviewing Vocabulary Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Compound Reviewing Vocabulary Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ionic Compound Reviewing Vocabulary Answers
 - Setting Reading Goals Ionic Compound Reviewing Vocabulary Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ionic Compound Reviewing Vocabulary Answers
 - Fact-Checking eBook Content of Ionic Compound Reviewing Vocabulary Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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