

# DIY Science Project Easy

## PERIODIC TABLE

The periodic table is displayed on a wooden frame with sticks. The elements are color-coded by groups:

- Alkali metals (green)
- Alkaline-earth metals (pink)
- Transition metals (blue)
- Other metals (orange)
- Other nonmetals (red)
- Halogens (green)
- Noble gases (pink)
- Rare-earth elements (yellow)
- Actinoid elements (purple)

The elements are arranged in rows and columns, with their atomic numbers and symbols labeled. The table includes elements from Hydrogen (1) to Oganesson (118).

Legend:

- Alkali metals
- Alkaline-earth metals
- Transition metals
- Other metals
- Other nonmetals
- Halogens
- Noble gases
- Rare-earth elements
- Actinoid elements

# PERIODIC TABLE

A hand-drawn periodic table of elements on a grid of 18 vertical sticks. The elements are represented by colored squares with their atomic number, symbol, and name. The layout follows the standard periodic table structure, including the s-block, p-block, d-block, and f-block (lanthanides and actinides).

| 1        | 2        | 3        | 4         | 5         | 6         | 7         | 8         | 9         | 10        | 11        | 12        | 13        | 14        | 15        | 16        | 17        | 18        |
|----------|----------|----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|
| 1<br>H   |          |          |           |           |           |           |           |           |           |           |           |           |           |           |           |           | 2<br>He   |
| 3<br>Li  | 4<br>Be  |          |           |           |           |           |           |           |           |           |           |           |           |           |           |           | 10<br>Ne  |
| 11<br>Na | 12<br>Mg |          |           |           |           |           |           |           |           |           |           |           |           |           |           |           | 18<br>Ar  |
| 19<br>K  | 20<br>Ca | 21<br>Sc | 22<br>Ti  | 23<br>V   | 24<br>Cr  | 25<br>Mn  | 26<br>Fe  | 27<br>Co  | 28<br>Ni  | 29<br>Cu  | 30<br>Zn  | 31<br>Ga  | 32<br>Ge  | 33<br>As  | 34<br>Se  | 35<br>Br  | 36<br>Kr  |
| 37<br>Rb | 38<br>Sr | 39<br>Y  | 40<br>Zr  | 41<br>Nb  | 42<br>Mo  | 43<br>Tc  | 44<br>Ru  | 45<br>Rh  | 46<br>Pd  | 47<br>Ag  | 48<br>Cd  | 49<br>In  | 50<br>Sn  | 51<br>Sb  | 52<br>Te  | 53<br>I   | 54<br>Xe  |
| 55<br>Cs | 56<br>Ba | 57<br>La | 72<br>Hf  | 73<br>Ta  | 74<br>W   | 75<br>Re  | 76<br>Os  | 77<br>Ir  | 78<br>Pt  | 79<br>Au  | 80<br>Hg  | 81<br>Tl  | 82<br>Pb  | 83<br>Bi  | 84<br>Po  | 85<br>At  | 86<br>Rn  |
| 87<br>Fr | 88<br>Ra | 89<br>Ac | 104<br>Rf | 105<br>Db | 106<br>Sg | 107<br>Bh | 108<br>Hs | 109<br>Mt | 110<br>Ds | 111<br>Rg | 112<br>Cn | 113<br>Nh | 114<br>Fl | 115<br>Mc | 116<br>Lv | 117<br>Ts | 118<br>Og |

|          |          |          |          |          |          |          |          |          |          |           |           |           |           |
|----------|----------|----------|----------|----------|----------|----------|----------|----------|----------|-----------|-----------|-----------|-----------|
| 58<br>Ce | 59<br>Pr | 60<br>Pm | 61<br>Nd | 62<br>Sm | 63<br>Eu | 64<br>Gd | 65<br>Tb | 66<br>Dy | 67<br>Ho | 68<br>Er  | 69<br>Tm  | 70<br>Yb  | 71<br>Lu  |
| 90<br>Th | 91<br>Pa | 92<br>U  | 93<br>Np | 94<br>Pu | 95<br>Am | 96<br>Cm | 97<br>Bk | 98<br>Cf | 99<br>Es | 100<br>Fm | 101<br>Md | 102<br>No | 103<br>Lr |

 **Animali non felici**

## Transition Metals

- other treatments

**Michelle Gauthier**

- Actinoid elements

● **hydrolysis** - splits molecules

Other people



**Pointe** — clearly is a long word.

# Making A Periodic Table Project

**Julia Schneider**



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