

Potentiometry

Choose the correct answer:

No	Sentence	answer
1	All of the following are related to reference electrode except: a) of known and constant potential b) obeys Nernst equation. c) of high resistance to permit current of electric current. d) insensitive to activity of analyte solution.	C
2	Potential of standard hydrogen electrode is: a) Zero b) related to H^+ concentration c) 0.059 PH d) A and B e) All of the above	d
3	The increase in chloride ion concentration in Ag/AgCl electrode leads to: a) increase the potential b) decrease the potential c) no effect d) none of them	b
4	Potential of calomel electrode is proportional to chloride ion concentration. a) directly b) inversely c) equal d) no relation	b
5	In silver/silver chloride electrode, as KCl concentration increases: a) silver ions increases, potential increases. b) silver ions decreases, potential increases. c) silver ions decreases, potential remains the same. d) silver ions decreases, potential decreases.	d
6	In calomel electrode, as KCl concentration increases: a) mercury(I) ions increases, potential increases. b) mercury(I) ions decreases, potential increases. c) mercury(I) ions decreases, potential remains the same. d) mercury(I) ions decreases, potential decreases.	d
7	What is the electrode potential for a half-cell consisting of a cadmium electrode immersed in a solution that is 0.0150 M in Cd^{2+} ($E^0 = -0.403$ V) a) 0.457 b) 0.403 c) -0.457 d) -0.201	C

Mcq On Potentiometry

Erzsébet Néher-Neumann



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Fouletier,Pierre Fabry,2010-01-26 This book reviews the state of art in the field of chemical sensors for analyses of ionic or molecular species dissolved in liquid media mainly in aqueous solutions The transduction of such devices is based on chemical biological and physical phenomena The fundamental phenomena involved in these sensors are described in the different chapters by specialists having a good expertise in the field Numerous recent bibliographic references are given Most of the devices could be miniaturised using modern technologies allowing a fabrication on a large scale for a mass

production at low cost Moreover such devices could open the field of applications in a near future environmental biomedical food industries domestic and automotive applications etc

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Advanced Potentiometry Erzsébet Néher-Neumann, 2010-03-14 Advanced Potentiometry details the accurate calculation of potentiometric titrations Additional terms such as the ideal diffusion potential together with the contribution of activity factors furnish calculated stability constants which do not include systematic errors and thus present a complete and correct description of equilibrium processes Moreover serious systematic errors resulting from the recent use of curve fitting computer programs are highlighted later in the text Advanced Potentiometry can advantageously be used by lecturers researchers PhD graduates and undergraduate students working in the fields of Solution Chemistry Physical Chemistry Inorganic Chemistry and Analytical Chemistry and Environmental Research

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