# NOMENCLATURE PACKET

Worksheet I: Binary Ionic Compounds (representative metals) - metals from groups 1A, 2A, and 3A (1, 2, and 13) have constant charges as ions and do NOT get Roman Numerals in their names

#### 1. Name the following ionic compounds:

a.	Al <sub>2</sub> O <sub>3</sub>	Aluminum oxide		
ь.	Cs2O	Cesium oxide		
c.	Rb <sub>3</sub> N	Rubidium nitride		
d.	Ca <sub>3</sub> N <sub>2</sub>	Calcium nitride		
c.	SrSe	Strontium selenide		
f.	Cs₂S	Cesium sulfide		
9.	Al <sub>2</sub> S <sub>3</sub>	Aluminum sulfide		
h.	LiBr	Lithium bromide		
i.	Mg <sub>3</sub> N <sub>2</sub>	Magnesium nitride		
i	CaF <sub>2</sub>	Calcium fluoride		

# 2. Write the chemical formula for the following ionic compounds:

a.	Barium nitride	Ba*2	N-3	Ba <sub>3</sub> N <sub>2</sub>
ь.	Indium Fluoride	In-3	F-1	InF <sub>3</sub>
<b>c</b> .	Calcium oxide	Ca-2	0-2	CaO
d.	Sodium nitride	Na <sup>-1</sup>	N-3	Na <sub>3</sub> N
e.	Magnesium chloride	Mg*2	CI-1	MgCl <sub>2</sub>
f.	Potassium oxide	K*1	0.2	K <sub>2</sub> O
9.	Magnesium oxide	Mg*2	0-2	MgO
h.	Potassium sulfide	K*1	5-2	K25
i	Lithium nitride	Li	N-3	Li <sub>3</sub> N
J-	Strontium fluoride	5r*2	F-1	SrF2
k.	Aluminum sulfide	A1-3	5-2	Al <sub>2</sub> S <sub>3</sub>
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# **Names And Formulas Worksheet 1 Answers**

**Suzanne Weixel, Michele Reader** 

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#### Table of Contents Names And Formulas Worksheet 1 Answers

- 1. Understanding the eBook Names And Formulas Worksheet 1 Answers
  - The Rise of Digital Reading Names And Formulas Worksheet 1 Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Names And Formulas Worksheet 1 Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Names And Formulas Worksheet 1 Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Names And Formulas Worksheet 1 Answers
  - Personalized Recommendations
  - Names And Formulas Worksheet 1 Answers User Reviews and Ratings
  - Names And Formulas Worksheet 1 Answers and Bestseller Lists
- 5. Accessing Names And Formulas Worksheet 1 Answers Free and Paid eBooks

- Names And Formulas Worksheet 1 Answers Public Domain eBooks
- Names And Formulas Worksheet 1 Answers eBook Subscription Services
- Names And Formulas Worksheet 1 Answers Budget-Friendly Options
- 6. Navigating Names And Formulas Worksheet 1 Answers eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Names And Formulas Worksheet 1 Answers Compatibility with Devices
  - Names And Formulas Worksheet 1 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Names And Formulas Worksheet 1 Answers
  - Highlighting and Note-Taking Names And Formulas Worksheet 1 Answers
  - Interactive Elements Names And Formulas Worksheet 1 Answers
- 8. Staying Engaged with Names And Formulas Worksheet 1 Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Names And Formulas Worksheet 1 Answers
- 9. Balancing eBooks and Physical Books Names And Formulas Worksheet 1 Answers
  - Benefits of a Digital Library
  - $\circ$  Creating a Diverse Reading Collection Names And Formulas Worksheet 1 Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Names And Formulas Worksheet 1 Answers
  - Setting Reading Goals Names And Formulas Worksheet 1 Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Names And Formulas Worksheet 1 Answers
  - Fact-Checking eBook Content of Names And Formulas Worksheet 1 Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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