

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \rightarrow 20 \rightarrow 10 \rightarrow 5 \rightarrow 2.5 \rightarrow 1.25$$

1 2 3 4 5 61.5 yr

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$100 \rightarrow 50 \rightarrow 25 \rightarrow 12.5 \text{ mg}$$

1 2 3 12.5 mg

3 half lives

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Modern Chemistry Answers Chapter Review

Ramananda Chatterjee



Modern Chemistry Answers Chapter Review:

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Principles of Modern Chemistry** David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses *PRINCIPLES OF MODERN CHEMISTRY* continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom *Modern Chemistry*, 1990 Books and Pamphlets, Including Serials and Contributions to Periodicals Library of Congress. Copyright Office, 1968 **The Modern Review** Ramananda Chatterjee, 1923 Includes section Reviews and notices of books **Introduction to Modern Chemistry: Student Manual** Eugene Meyer, 1979 Catalog of Copyright Entries. Third Series Library of Congress. Copyright Office, 1968 Barron's how to Prepare for College Entrance Examinations Samuel C. Brownstein, Mitchel Weiner, 1974 A guide to preparing for college entrance examinations with emphasis on study programs for the verbal mathematics and standard written English parts of the SAT Includes practice tests **Nursing School Entrance Exams** Sandra S. Swick, Rita R. Callahan, 2020-06-19 Barron's Nursing School Entrance Exams provides detailed review and practice materials that you need to achieve success on the various Nursing School Entrance Exams including the HESI A2 NLN PAX RN PSB RN RNEE and the TEAS This edition features A multi part exam that covers all of the topic areas and question types seen on most nursing school entrance exams A diagnostic test so you can assess your strengths and weaknesses in each topic area before beginning your review Comprehensive review and practice material for all Verbal Ability Reading Comprehension and Numerical Ability topics An entire review and practice section for all Science topics with each section broken down into an outline format for quick studying and sample tests for every topic Test taking strategies and answers to frequently asked questions about preparing for your entrance exam Strategies for answering each question type You ll also get information about nursing programs and the profession in general Curriculum Review, 1986 The Electrical Review, 1928 Academy; a Weekly Review of Literature, Learning, Science and Art, 1882 The Poetical gazette the official organ of the Poetry society and a review of poetical affairs nos 4 7 issued as supplements to the Academy v 79 Oct 15 Nov 5 Dec 3 and 31 1910 *Engineering and Boiler House Review*, 1917 The Siebel Technical Review, 1928 *Academy, with which are Incorporated Literature and the English Review*, 1882 *Catholic Educational Review* Edward Aloysius Pace, Thomas

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