# Names & Formulas for Ionic Compounds

Give the name or formula of the following ionic compounds:

	Name		Formula
1)	Na <sub>2</sub> CO <sub>3</sub>	21)	sodium phosphide
2)	NaOH	22)	magnesium nitrate
3)	MgBr <sub>2</sub>	23)	lead (II) sulfite
1)	ксі	24)	calcium phosphate
5)	FeCl <sub>2</sub>	25)	ammonium sulfate
3)	FeCb	26)	silver cyanide
7)	Zn(OH)2	27)	aluminum sulfide
3)	Be <sub>2</sub> SO <sub>4</sub>	28)	beryllium chloride
9)	CrF <sub>2</sub>	29)	copper (I) arsenide
0)	Al <sub>2</sub> S <sub>3</sub>	30)	iron (III) oxide
1)	Рьо	31)	gallium nitride
2)	Li <sub>3</sub> PO <sub>4</sub>	32)	iron (II) bromide
3)	Till4	33)	vanadium (V) phosphate
4)	Co3N2	34)	calcium oxide
15)	Mg <sub>3</sub> P <sub>2</sub>	35)	magnesium acetate
6)	Ga(NO <sub>2</sub> ) <sub>3</sub>	36)	aluminum sulfate
7)	Ag <sub>2</sub> SO <sub>3</sub>	37)	copper (I) carbonate
8)	NH4OH	38)	barium oxide
19)	Al(CN)3	39)	ammonium sulfite
20)	Be(CH <sub>3</sub> COO) <sub>2</sub>	40)	silver bromide

# **Names And Formulas Worksheet 1**

**RJ Shavelson** 

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