



Nomenclature for polyatomic ions Worksheet-Answer Key

Exercise 1. Complete the table of neutral ionic compounds with the formulas and names for each cation-anion pair, similar to Exercise 2.

	NH_4^+	Na^+	PO_4^{3-}	CO_3^{2-}	ClO_2^-	OH^-
Na^+	Na_2SO_4 sodium sulfate	NaNO_3 sodium nitrate	Na_3PO_4 sodium phosphate	Na_2CO_3 sodium carbonate	NaClO_2 sodium chlorite	NaOH sodium hydroxide
Al^{3+}	$\text{Al}_2(\text{SO}_4)_3$ aluminum sulfate	AlNO_3 aluminum nitrate	AlPO_4 aluminum phosphate	$\text{Al}_2(\text{CO}_3)_3$ aluminum carbonate	AlClO_2 aluminum chlorite	AlOH aluminum hydroxide
Ba^{2+}	BaSO_4 barium sulfate	$\text{Ba}(\text{NO}_3)_2$ barium nitrate	$\text{Ba}_3(\text{PO}_4)_2$ barium phosphate	BaCO_3 barium carbonate	BaClO_2 barium chlorite	BaOH barium hydroxide
NH_4^+	$(\text{NH}_4)_2\text{SO}_4$ ammonium sulfate	NH_4NO_3 ammonium nitrate	$(\text{NH}_4)_3\text{PO}_4$ ammonium phosphate	$(\text{NH}_4)_2\text{CO}_3$ ammonium carbonate	NH_4ClO_2 ammonium chlorite	NH_4OH ammonium hydroxide
Ca^{2+}	Ca_2SO_4 calcium sulfate	CaNO_3 calcium nitrate	Ca_3PO_4 calcium phosphate	Ca_2CO_3 calcium carbonate	CaClO_2 calcium chlorite	CaOH calcium hydroxide

Naming Polyatomic Ions Packet Answer Key

A Gutmann



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