

# Honors Chemistry Final Exam Study Guide

## 2024 with Complete Solutions

**Mole** - Answer- SI unit for measuring the amount of substance;

**Molar Mass** - Answer- the mass of a mole of an element

**Representative particles** - Answer- refers to the species present in a substance: usually atoms, molecules, or formula units

**Avogadro's number** - Answer- the number of representative particles in a mole

**Avogadro's hypothesis** - Answer- states that equal volumes of gases at the same temperature and pressure contain equal numbers of particles

**Molar volume** - Answer- the quantity, 22.4 L

**Dimensional analysis** - Answer- mole to mass, mole to volume

**Percent Composition** - Answer- percent by mass of each element in the compound by which the relative amounts of the elements in a compound are expressed

**STP (Standard Temperature and Pressure)** - Answer- a temperature of 0°C and a pressure of 101.3 kPa, or 1 atmosphere (atm)

**Empirical formula** - Answer- the basic ratio gives the lowest whole-number ratio of the atoms of the elements contained in the compound

**Molecular formula** - Answer- either the same as its experimentally determined empirical formula, or it is a simple whole-number multiple of its empirical formula

**Types of reactions** - Answer- the five general types of reaction are combination, decomposition, single-replacement, double-replacement, and combustion

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