

6. Polystyrene	Styrene	$\text{CH}_2=\text{CH}$ C_6H_5
7. Buna rubber	Buta-1,3-diene	$\text{CH}_2=\text{CH}-\text{CH}=\text{CH}_2$
8. Dacron (Terylene)	Ethane-1,2-diol + 1,4-benzenedioic acid	CH_2-CH_2 OH OH + $\text{HO}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{C}_6\text{H}_4)-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$
9. Nylon 6,6	1,6-diamino hexane + 1,6-hexanedioic acid	$\text{NH}_2-(\text{CH}_2)_6-\text{NH}_2$ + $\text{HO}-\overset{\text{O}}{\parallel}{\text{C}}-(\text{CH}_2)_4-\overset{\text{O}}{\parallel}{\text{C}}-\text{OH}$ $\text{C}_6\text{H}_5-\text{OH}$ + HCHO
10. Bakelite	Phenol + formaldehyde	

2. On the basis of monomers.

a. Homopolymers

→ It is formed by only one type of monomer unit.

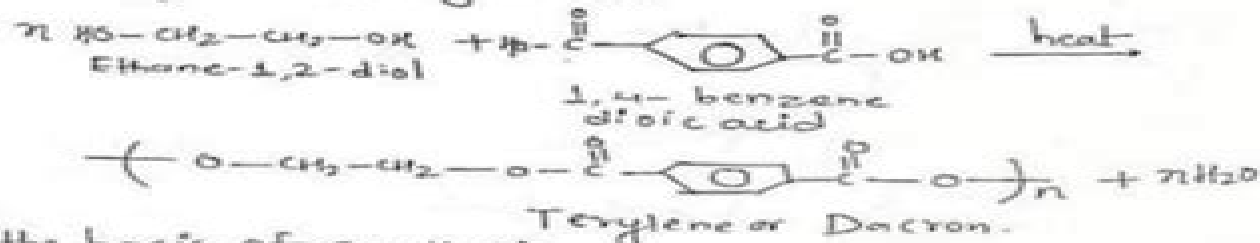
For eg:- polythene, polyvinyl chloride (PVC), Orlon, Teflon etc.

b. Copolymers.

→ It is formed by more than one type of monomers

→ Monomers are arranged in definite structural pattern in a random manner.

For eg Dacron, Nylon 6,6



3 On the basis of synthesis.

a. Addition polymers

- Large number of monomer molecules are added together to form high molecular weight.

For eg:- pvc, orlon, te flon

Mcq In Nuclear Chemistry

Peter A C Mcpherson



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