Borazine, also known as inorganic benzene, can be prepared by the reaction of 3-equivalents of "X" with 6-equivalents of "Y". "X" and "Y", respectively are :

 $\bigcirc$  B(OH)<sub>3</sub> and NH<sub>3</sub>

 $egin{array}{c} egin{array}{c} \egin{array}{c} \egin{array}{c} \egin{array}{c} \egin{array}{c} \egin{array}{c} \egin{array}$ 

 $\bigcirc$   $\mathbf{B_2H_6}$  and  $\mathbf{HN_3}$ 

 $\bigcirc$  NH $_3$  and B $_2$ O $_3$ 

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# **SAYMOUR SHAPIRO**

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