

Q.No.01 Choose the correct answer:

1. The average value of physical quantity $A(q, p)$ for actual system of interest is

- a) $\bar{A} = \frac{1}{\Omega} \int_{-\infty}^{+\infty} A(q, p) P(q, p, t) dt$ b) $\bar{A} = \frac{1}{\Omega} \int_{-\infty}^{+\infty} A(q, p) \rho(q, p, t) dt$ c) $1 = \frac{1}{\Omega} \int_{-\infty}^{+\infty} A(q, p) \rho(q, p, t) dt$ d) $\bar{A} = \frac{1}{\Omega} \int_{-\infty}^{+\infty} A(q, p) U(q, p, t) dt$

2. The value of $\gamma(C_p/C_v)$ for a DIATOMIC GAS molecule is

- a) 1.66 b) 1.4 c) 1.33 d) 1

The trajectory of S.H.O in phase space is

- a) Ellipse b) Circle c) Parabola d) Spiral

4. In phase space, the total probability is obtained by taking

- a) Sum of individual probability b) Product of individual probability c) Subtraction of individual probability d) Division of individual probability

5. According to Maxwell-Boltzmann velocity distribution; the average speed $\bar{u}$ of the particle of a classical gas is

- a) $8k_B T / \pi m$ b) $8k_B T / (\pi m)^{1/2}$ c) $(\pi m / 8k_B T)^{1/2}$ d) $\pi m / 8k_B T$

6. If a change is reversible, we cannot have

- a) $\Delta S_{\text{universe}} > 0$ b) $\Delta S_{\text{universe}} = 0$ c) $\Delta S_{\text{universe}} < 0$ d) both a) & b)

7. PLANK's formula for BLACK BODY radiation can be derived from STATISTICS

- a) Fermi-Dirac b) Bose-Einstein c) Max Well-Boltzmann d) (a) and (b)

8. If the wave function of identical particles change sign when these are interchanged, the particles are

- a) Boson b) Fermions c) Photons d) Negatron

9. In case of micro canonical ensemble, the volume of sphere and its thickness should be

- a) Greater than 1 b) Less than 0 c) Different d) Same

10. Which of the following is true in regard to the energy of an isolated system

- a) $dQ \neq 0$ b) $dW \neq 0$ c) $E = \text{constant}$ d) All of the mentioned

11. A red glass piece is heated until it glows in dark. The color of the glow will be

- a) Red b) Orange c) Green d) Violet

12. Which of the following parameters does not characterize the thermodynamic state of matter?

- a) Volume b) Temperature c) Pressure d) Work

13. Which is meaningful arrangements for case of n particles and two boxes, if n_1 in box-1 and remaining $n_2 = n - n_1$ in box-2

- a) $n! \cdot n_2! / n_1!$ b) $n! / n_1! \cdot n_2!$ c) $n! \cdot (n! - n_1!) / n_1!$ d) $n! \cdot (n! - n_1!) / n_2!$

14. What will the total number of microstates for particles $n = 3$ and compartments $C = 2$

- a) 9 b) 5 c) 8 d) 1

15. The relation for heat conductivity of monatomic gases is (where, l is mean free path, ρN is particle density)

- a) $\lambda = 3/2 k_B v \rho N$ b) $\lambda = 1/3 k_B v \rho N$ c) $\lambda = 2/3 k_B v \rho N$ d) $\lambda = 1/2 k_B v \rho N$

16. Temperature T and Chemical potential μ may be used to characterize an

- a) Closed system b) Particle system c) Open system d) Isolated system

17. The probability of occurrence of two independent events is equal to their

- a) Sum b) Difference c) product d) Ratio

18. The system will do work as much its Helmholtz free energy as

- a) $F = P \log Z$ b) $F = -T \log Z$ c) $F = T \log Z$ d) $F = -T \log 4$

19. The difference between molar specific heats at constant pressure and at constant volume is always equal to about

- a) Avogadro's number b) Gas constant c) 1.41 d) Gibbs Constant

20. The average number of particles (probability) in Energy state ϵ_k is given as

- a) $1 / 1 + [\exp (\epsilon - \mu) / T]$ b) $1 / 1 - [\exp (\epsilon - \mu) / T]$ c) $1 / 1 + [\exp (\epsilon + \mu) / T]$ d) $1 / 1 + [\exp (\epsilon - \mu)]$

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