

Name: _____

SHORT ANSWER. Write the word or phrase that best completes each statement or answers the question.

- 1) What is the solvent in an aqueous solution?

ESSAY. Write your answer in the space provided or on a separate sheet of paper.

- 2) Explain why water is a good solvent for ionic substances.

MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 3) Of the species below, only _____ is not an electrolyte.
A) HCl B) Ar C) NaCl D) KOH E) Rb_2SO_4
- 4) What is the molarity of an aqueous solution containing 75.3 g of glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) in 35.5 mL of solution?
A) 0.197 B) 2.12 C) 3.52 D) 1.85 E) 11.8
- 5) The concentration of species in 500 mL of a 2.104 M solution of sodium sulfate is _____ M sodium ion and _____ M sulfate ion.
A) 4.208, 2.104 B) 1.052, 1.052 C) 2.104, 1.052 D) 2.104, 4.208 E) 2.104, 2.104
- 6) What is the molarity of a solution prepared by diluting 43.72 mL of 1.005 M aqueous $\text{K}_2\text{Cr}_2\text{O}_7$ to 500 mL?
A) 0.870 B) 87.9 C) 0.0879 D) 0.0115 E) 0.0218
- 7) When aqueous solutions of _____ are mixed, a precipitate forms.
A) NaI and KBr
B) Li_2CO_3 and CsI
C) K_2SO_4 and CrCl_3
D) NiBr_2 and AgNO_3
E) KOH and $\text{Ba}(\text{NO}_3)_2$
- 8) What are the spectator ions in the reaction between KOH (aq) and HNO_3 (aq)?
A) H^+ and NO_3^-
B) OH^- only
C) K^+ and NO_3^-
D) K^+ and H^+
E) H^+ and OH^-

SHORT ANSWER. Write the word or phrase that best completes each statement or answers the question.

- 9) What is the concentration of HCl in a 0.100 M solution of HCl? Why?

Multiple Choice Ap Chemistry Practice Test 1

A Gutmann



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Table of Contents Multiple Choice Ap Chemistry Practice Test 1

1. Understanding the eBook Multiple Choice Ap Chemistry Practice Test 1
 - The Rise of Digital Reading Multiple Choice Ap Chemistry Practice Test 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice Ap Chemistry Practice Test 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Multiple Choice Ap Chemistry Practice Test 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice Ap Chemistry Practice Test 1
 - Personalized Recommendations
 - Multiple Choice Ap Chemistry Practice Test 1 User Reviews and Ratings
 - Multiple Choice Ap Chemistry Practice Test 1 and Bestseller Lists

5. Accessing Multiple Choice Ap Chemistry Practice Test 1 Free and Paid eBooks
 - Multiple Choice Ap Chemistry Practice Test 1 Public Domain eBooks
 - Multiple Choice Ap Chemistry Practice Test 1 eBook Subscription Services
 - Multiple Choice Ap Chemistry Practice Test 1 Budget-Friendly Options
6. Navigating Multiple Choice Ap Chemistry Practice Test 1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Multiple Choice Ap Chemistry Practice Test 1 Compatibility with Devices
 - Multiple Choice Ap Chemistry Practice Test 1 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Multiple Choice Ap Chemistry Practice Test 1
 - Highlighting and Note-Taking Multiple Choice Ap Chemistry Practice Test 1
 - Interactive Elements Multiple Choice Ap Chemistry Practice Test 1
8. Staying Engaged with Multiple Choice Ap Chemistry Practice Test 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Multiple Choice Ap Chemistry Practice Test 1
9. Balancing eBooks and Physical Books Multiple Choice Ap Chemistry Practice Test 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Multiple Choice Ap Chemistry Practice Test 1
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Multiple Choice Ap Chemistry Practice Test 1
 - Setting Reading Goals Multiple Choice Ap Chemistry Practice Test 1
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Multiple Choice Ap Chemistry Practice Test 1
 - Fact-Checking eBook Content of Multiple Choice Ap Chemistry Practice Test 1
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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