

Stoichiometry: Multiple Choice

1. The limiting reactant of a chemical reaction is

- A) the reactant of a chemical process that is not consumed completely
- B) the product of a chemical process that is consumed completely
- C) the product of a chemical process that is not consumed completely
- D) the reactant of a chemical process that is consumed completely
- E) none of the above

ANS: D

1. Consider the following reaction; $\text{Ca}(\text{OH})_{2(aq)} + \text{Al}_2(\text{SO}_4)_{3(aq)} \rightarrow \text{CaSO}_{4(s)} + \text{Al}(\text{OH})_{3(s)}$

When balanced, the mole ratio for calcium hydroxide reacting with aluminum sulphate is

- A) 1:1
- B) 2:1
- C) 2:3
- D) 1:2
- E) 3:1

ANS: E

1. Sodium phosphate, Na_3PO_4 , can be prepared by combining sodium hydroxide and hydrogen phosphate as shown in the following balanced chemical equation:



If 2.5 mol of H_2PO_4 are reacted, the number of moles of water that will be produced is

- A) 1.0
- B) 2.5
- C) 3.0
- D) 5.0
- E) 7.5

ANS: E

1. Potassium chloride and oxygen gas are formed upon heating potassium chlorate. The following balanced equation shows this decomposition process:



The number of moles of KClO_3 required to produce 5.0 mol of oxygen is

- A) 1.0
- B) 2.0
- C) 3.0
- D) 3.3
- E) 5.0

ANS: D

1. Determine the number of grams of CO_2 that can be produced from 48.0 g of the hydrocarbon C_3H_8 , according to the following equation:



- A) 37.6 g
- B) 50.2 g
- C) 87.8 g
- D) 151 g
- E) 188.4 g

1. The limiting reactant in a chemical reaction is the _____ that is _____.

- A) Reactant, in excess
- B) Product, left over
- C) Reactant, used up
- D) Product, used up

ANS: C

Multiple Choice Questions On Stoichiometry

Arshad Iqbal



Multiple Choice Questions On Stoichiometry:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Multiple Choice Questions On Stoichiometry**

In a global inundated with monitors and the cacophony of instant transmission, the profound power and emotional resonance of verbal art often fade in to obscurity, eclipsed by the continuous barrage of sound and distractions. However, situated within the musical pages of **Multiple Choice Questions On Stoichiometry**, a fascinating work of literary elegance that pulses with organic feelings, lies an memorable journey waiting to be embarked upon. Published with a virtuoso wordsmith, this mesmerizing opus manuals viewers on a psychological odyssey, delicately exposing the latent possible and profound influence embedded within the intricate internet of language. Within the heart-wrenching expanse with this evocative evaluation, we can embark upon an introspective exploration of the book is main themes, dissect its captivating writing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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