

SOLUTIONS OF ACIDS AND BASES

- ① Write the chemical equation showing how each of the following dissolves in water.



- ② For each reaction type, complete the equations and balance them.

a. Making acids from nonmetal oxide

General form: nonmetal oxide + water \longrightarrow acid

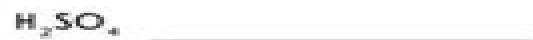


b. Making bases from metal oxide

General form: Metal oxide + water \longrightarrow base



- ③ Write the balanced equation for the titration between sulfurous acid and potassium hydroxide.
- ④ Write the balanced net ionic equation for the titration between sulfurous acid and potassium hydroxide.
- ⑤ Write an equation for the reaction of potassium metal with hydrochloric acid.
- ⑥ Acids can be labelled as monoprotic, diprotic, or triprotic. Label the following as monoprotic, diprotic, or triprotic



Lecture 30 Acids And Bases Solutions Worksheet

Apchem

Rachel S Tattersall



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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Lecture 30 Acids And Bases Solutions Worksheet Apchem**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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