

Balancing Chemical Equations

Balance the equations below:

- 1) $\underline{1} \text{ N}_2 + \underline{3} \text{ H}_2 \rightarrow \underline{2} \text{ NH}_3$
- 2) $\underline{2} \text{ KClO}_3 \rightarrow \underline{2} \text{ KCl} + \underline{3} \text{ O}_2$
- 3) $\underline{2} \text{ NaCl} + \underline{1} \text{ F}_2 \rightarrow \underline{2} \text{ NaF} + \underline{1} \text{ Cl}_2$
- 4) $\underline{2} \text{ H}_2 + \underline{1} \text{ O}_2 \rightarrow \underline{2} \text{ H}_2\text{O}$
- 5) $\underline{1} \text{ Pb(OH)}_2 + \underline{2} \text{ HCl} \rightarrow \underline{2} \text{ H}_2\text{O} + \underline{1} \text{ PbCl}_2$
- 6) $\underline{2} \text{ AlBr}_3 + \underline{3} \text{ K}_2\text{SO}_4 \rightarrow \underline{6} \text{ KBr} + \underline{1} \text{ Al}_2(\text{SO}_4)_3$
- 7) $\underline{1} \text{ CH}_4 + \underline{2} \text{ O}_2 \rightarrow \underline{1} \text{ CO}_2 + \underline{2} \text{ H}_2\text{O}$
- 8) $\underline{1} \text{ C}_3\text{H}_8 + \underline{5} \text{ O}_2 \rightarrow \underline{3} \text{ CO}_2 + \underline{4} \text{ H}_2\text{O}$
- 9) $\underline{2} \text{ C}_8\text{H}_{18} + \underline{25} \text{ O}_2 \rightarrow \underline{16} \text{ CO}_2 + \underline{18} \text{ H}_2\text{O}$
- 10) $\underline{1} \text{ FeCl}_3 + \underline{3} \text{ NaOH} \rightarrow \underline{1} \text{ Fe(OH)}_3 + \underline{3} \text{ NaCl}$
- 11) $\underline{4} \text{ P} + \underline{5} \text{ O}_2 \rightarrow \underline{2} \text{ P}_2\text{O}_5$
- 12) $\underline{2} \text{ Na} + \underline{2} \text{ H}_2\text{O} \rightarrow \underline{2} \text{ NaOH} + \underline{1} \text{ H}_2$
- 13) $\underline{2} \text{ Ag}_2\text{O} \rightarrow \underline{4} \text{ Ag} + \underline{1} \text{ O}_2$
- 14) $\underline{1} \text{ S}_8 + \underline{12} \text{ O}_2 \rightarrow \underline{8} \text{ SO}_2$
- 15) $\underline{6} \text{ CO}_2 + \underline{6} \text{ H}_2\text{O} \rightarrow \underline{1} \text{ C}_6\text{H}_{12}\text{O}_6 + \underline{6} \text{ O}_2$
- 16) $\underline{2} \text{ K} + \underline{1} \text{ MgBr}_2 \rightarrow \underline{2} \text{ KBr} + \underline{1} \text{ Mg}$
- 17) $\underline{2} \text{ HCl} + \underline{1} \text{ CaCO}_3 \rightarrow \underline{1} \text{ CaCl}_2 + \underline{1} \text{ H}_2\text{O} + \underline{1} \text{ CO}_2$
- 18) $\underline{1} \text{ HNO}_3 + \underline{1} \text{ NaHCO}_3 \rightarrow \underline{1} \text{ NaNO}_3 + \underline{1} \text{ H}_2\text{O} + \underline{1} \text{ CO}_2$
- 19) $\underline{2} \text{ H}_2\text{O} + \underline{1} \text{ O}_2 \rightarrow \underline{2} \text{ H}_2\text{O}_2$
- 20) $\underline{2} \text{ NaBr} + \underline{1} \text{ CaF}_2 \rightarrow \underline{2} \text{ NaF} + \underline{1} \text{ CaBr}_2$
- 21) $\underline{1} \text{ H}_2\text{SO}_4 + \underline{2} \text{ NaNO}_2 \rightarrow \underline{2} \text{ HNO}_2 + \underline{1} \text{ Na}_2\text{SO}_4$

$\text{Mg}^{2+} \text{ Br}^-$
~~ERROR~~
 $\text{MgBr}_2 \rightarrow$

L Farrell Balancing Chemical Equations Answers

L Reisser



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Embracing the Song of Expression: An Psychological Symphony within **L Farrell Balancing Chemical Equations Answers**

In some sort of eaten by displays and the ceaseless chatter of immediate conversation, the melodic elegance and psychological symphony created by the prepared term frequently diminish in to the back ground, eclipsed by the persistent noise and disruptions that permeate our lives. But, nestled within the pages of **L Farrell Balancing Chemical Equations Answers** a wonderful fictional value filled with raw thoughts, lies an immersive symphony waiting to be embraced. Constructed by a masterful composer of language, that charming masterpiece conducts viewers on an emotional journey, well unraveling the hidden melodies and profound influence resonating within each carefully constructed phrase. Within the depths of this emotional evaluation, we shall discover the book is key harmonies, analyze their enthralling publishing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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