#### May 2009 TZ1 Question 1

(b) For part of her extended essay investigation into the efficiency of the process, a student reacted a pure sample of a vegetable oil (where R=C<sub>1</sub>,H<sub>20</sub>) with methanol. The raw data recorded for the reaction is below.

Mass of oil	100	1013.0 g
and and the control of the control o	-	200.0 g
Mass of sodium hydroxide	-	3.5 g
Mass of biodiesel produced	1000	811.0 g

(in moles) of the oil and the methanol used, and hence the amount (in moles) of excess methanol.

The relative molecular mass of the oil used by the student is 885.6. Calculate the amount

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#### May 2009 TZ2 Question 1

(b) A student reacted some salicylic acid with excess ethanoic anhydride. Impure solid aspirin was obtained by filtering the reaction mixture. Pure aspirin was obtained by recrystallization. The following table shows the data recorded by the student.

Mass of salicylic acid used	$3.15 \pm 0.02 \text{ g}$
Mass of pure aspirin obtained	2.50 ± 0.02 g

- (i) Determine the amount, in mol, of salicylic acid, C<sub>s</sub>H<sub>s</sub>(OH)COOH, used. [2]
- (ii) Calculate the theoretical yield, in g, of aspirin, C<sub>s</sub>H<sub>s</sub>(OCOCH<sub>s</sub>)COOH. /2/

# **May 2013 Chemistry Sl Paper 1**

**Guillaume Favre** 

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**Production Chemicals for the Oil and Gas Industry** Malcolm A. Kelland, 2014-03-13 This text discusses a wide variety of production chemicals used by the oil and gas industry for down hole and topside applications both onshore and offshore It reviews all past and present classes of production chemicals providing numerous difficult to obtain references Unlike other texts that focus on how products perform in the field this book focuses on the specific structures of chemicals that are known to deliver the required or desired performance Where known it also details the environmental aspects of the chemicals discussed and their success in the field Flow Assurance in Pipelines Saeid Mokhatab, Juan Manzano-Ruiz, Antonin Chapoy, Jonathan J. Wylde, 2025-11-27 Flow assurance is critical for effective design and operation of hydrocarbon production and transmission systems The aim is to ensure safe and economical flow of hydrocarbon fluids from the source to the markets As future development of hydrocarbon resources moves into more challenging environments the importance of flow assurance is becoming more asserted Written by an internationally recognized team of experts Flow Assurance in Pipelines A Reference Guide is a unique well researched and comprehensive work in the flow assurance discipline that addresses the stages of a flow assurance study and identifies how they relate to the pipeline system design and operation This book is a fundamental resource for flow assurance engineers pipeline engineers operators and production chemists providing a detailed explanation of basic flow assurance issues that can arise in the pipeline systems and discusses practical methods to solve those issues safely and cost effectively Provides a systematic understanding of the concept of flow assurance Discusses flow assurance challenges in pipelines and how to overcome them Reviews latest flow assurance technologies engineering principles and practices Describes the best operating philosophies of pipeline systems to minimize flow assurance risks Explores sustainable pipeline applications in the transition to a low carbon economy Catalog of Instructional Films (16mm, Super 8, and 8mm) for College Chemistry Advisory Council on College Chemistry, 1969 Medical Aspects of Biological Warfare, 2e Joel Bozue, Christopher K. Cote, Pamela J. Glass, The Borden Institute, U.S. Army Medical Department, 2018-08-01 Reflecting the critical threat posed by biological warfare and terrorism in a post 9 11 world Medical Aspects of Biological Warfare 2e addresses the weaponization of biological agents categorizing potential agents as food waterborne or agricultural agents or toxins and discusses their respective epidemiology Recent advances in biomedical knowledge are presented that include descriptions of individual agents and the illnesses induced Authors discuss biotoxins and explain methods for early identification for anthrax plague smallpox alphaviruses and staphylococcal enterotoxins Case studies and research on successful management practices treatments and antidotes are also included Contains updated and revised material since previous 2007 edition Previous Print Hardcover ISBN 9780160797316 eBook 9780160872389 Related products More published products by The Borden Institute U S Army Medical Department AMEDD are here https bookstore gpo gov agency army medical department amedd Arms Weapons collection is available here https bookstore gpo gov catalog

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web for this experiment your students will explore basic chemistry concepts by testing the solubility of different substances in water from the example above we know that table salt is highly soluble in water

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web the way that matter behaves is called a property of matter one important property of matter is called solubility we think about solubility when we dissolve something in water or another liquid if a chemical is soluble in water then the chemical will dissolve or appear to vanish when you add it to water

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web have different solubilities in middle school students can weigh the solutes for a solubility test that uses equal masses 2 add 10 milliliters 2 teaspoons of water to two separate cups 3 at the same time pour the water into the sugar and salt cups 4

gently swirl the cups to see whether sugar or salt dissolves the most expected results

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web dissolving a solution is made when one substance called the solute dissolves into another substance called the solvent dissolving is when the solute breaks up from a larger crystal of molecules into much smaller groups or individual molecules this break up is caused by coming into contact with the solvent

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web this type of experiment introduces aqueous solutions or solutions of a substance dissolved in water to students the experiment can also spark a discussion about why water is able to

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web average 6 10 days credits note for this science project you will need to develop your own experimental procedure use the information in the summary tab as a starting place if you would like to discuss your ideas or need help troubleshooting use the ask an expert forum our experts won t do the work for you but they will make

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web use these fully differentiated resources to help your pupils understand the concept of solubility and to plan a school science experiment support them through their planning and write up using the lower and higher ability write up sheets encourage them to write a plan collect results write a conclusion and then finally evaluate the investigation they 13 lessons to teach about the chemistry of mixtures and solutions - Jul 02 2022

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out if there is a limit to how much solute can be added to the solvent when making a solution

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web to make sure that your results are reproducible you should repeat your solubility experiment at least three separate times for each chemical for each solubility determined by method 1 you will have the original volume of water the total mass of the solute and the remaining mass of the solute

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create a solubility curve by the end of this lab students should be able to properly use an analytical balance to measure mass solutions solubility and miscibility mnstep activity mini collection - Sep 04 2022

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