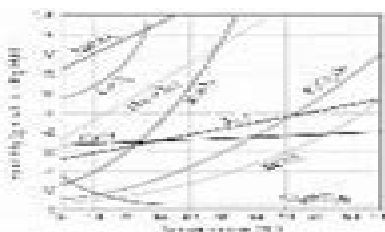


1. A sample of potassium nitrate (49.0 g) is dissolved in 101 g of water at 100 °C with precautions taken to avoid evaporation of any water. The solution is cooled to 30.0 °C and a small amount of precipitate is observed. This solution is

- A) hydrated B) placated C) saturated D) unsaturated E) supersaturated

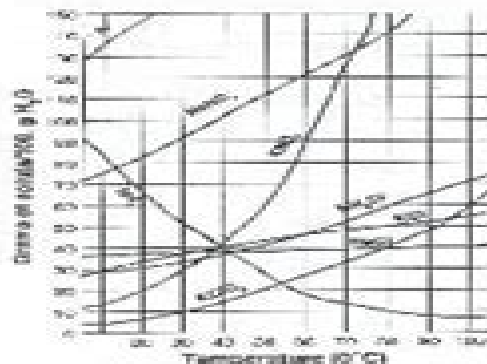
ANS: C



1. The solubility of MnSO_4 monohydrate in water at 20 °C is 70.0 g per 100.0 mL of water. A solution at 20 °C that is 4.22 M in MnSO_4 monohydrate is best described as a(n) _____ solution. The formula weight of MnSO_4 monohydrate is 168.97 g/mol.

- A) hydrated B) solvated C) saturated D) unsaturated E) supersaturated

ANS: E



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