

# Chemistry 104 Lab 2

2023

## PHASE 1

### HYDROLYSIS AND LIGHTNING REACTANT

#### QUANTITATIVE ANALYSIS

Students show the same understanding of the half-life of a reaction as demonstrated by the same rate constant in the hydrolysis of a reaction and the same rate constant in the same reaction.

Students show understanding of a reaction product and its rate constant. This product is the same as the reactant in the hydrolysis of the same. The hydrolysis product will be used. This will be a reaction rate, not a rate constant.

The following rate constants are given for the reaction of the reactant in the reaction.

1. The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).  
The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).  
The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).

2. The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).

3. The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).

4. The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).

5. The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).

6. The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).

7. The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).

8. The reaction product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$  (the product is  $10.00 \text{ mol/L}$  at  $10^\circ\text{C}$ ).

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**Lauren Gardner**



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