

POGIL 2 Ions

How are ions made from neutral atoms?

You have learned that not all atoms of an element are the same. Variation in the number of neutrons results in different isotopes of the element. In this activity we will explore another variation that can take place—the loss and gain of electrons. The exchange of electrons between atoms is a very common way for chemical change to take place. We will see it many times throughout the year.

1. Use Diagram 1 on the next page to complete the following table.

	Metal or Nonmetal	Number of protons in the atom	Number of protons in the ion	Number of neutrons in the atom	Number of neutrons in the ion	Number of electrons in the atom	Number of electrons in the ion	Charge on the ion
Lithium	metal	3	3	4	4	3	2	1+
Magnesium	<i>Metal</i>	12	12	12	12	12	10	2+
Aluminum	<i>Metalloid</i>	13	13	14	14	13	10	3+
Fluorine	<i>Non metal</i>	9	9	10	10	9	10	1-
Oxygen	nonmetal	8	8	8	8	8	10	2-
Nitrogen	<i>Non metal</i>	7	7	7	7	7	10	3-

2. Based on the table you completed in Question 1, what distinguishes a neutral atom from an ion?

A neutral atom has the same number of protons and electrons (no charge). An ion has a charge (# of protons does not equal # of electrons).

3. Examine the isotope symbols in Diagram 1.

- a. Where is the ion charge located in the isotope symbol?

Top right corner

- b. Is a charge indicated on the neutral atoms? If yes, where is it located?

No

4. Which subatomic particle carries a positive charge?

Protons are positive

5. Which subatomic particle carries a negative charge?

Electrons are negative

6. Propose a mathematical equation to calculate the charge on an ion from the number of protons and electrons in an ion. Confirm that your equation works using two positive ion examples and two negative ion examples from Diagram 1.

The charge of an ion = # of protons - # of electrons

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Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more

Chemical Pedagogy Keith S Taber, 2024-12-20 How should chemistry be taught in schools colleges and universities Chemical Pedagogy discusses teaching approaches and techniques the reasoning behind them and the evidence for their effectiveness The book surveys a wide range of different pedagogic strategies and tactics that have been recommended to better engage learners and provide more effective chemistry teaching These accounts are supported by an initial introduction to some key ideas and debates about pedagogy the science of teaching Chemical Pedagogy discusses how teaching innovations can be tested to inform research based practice Through this book the author explores the challenges of carrying out valid experimental studies in education and the impediments to generalising study results to diverse teaching and learning contexts As a result the author highlights both the need to read published studies critically and the value of teachers and lecturers testing out recommended innovations in their own classrooms Chemical Pedagogy introduces core principles from research into human cognition and learning to provide a theoretical perspective on how to best teach for engagement and understanding An examination of some of the more contentious debates about pedagogy leads to the advice to seek optimally guided instruction which balances the challenge offered to learners with the level of support provided This provides a framework for discussing a wide range of teaching approaches and techniques that have been recommended to those teaching chemistry across educational levels including both those intended to replace teaching from the front and others that can be built into traditional lecture courses to enhance the learning experience

Ionic Processes in Solution Ronald Wilfrid Gurney, 1962

Ions in Solution (1) (sound Recording). John Burgess, 1984

Ion Solvation Y. Marcus, 1985 Chemical reactions generally take place in solution and often involve ions The behaviour of ions in solution manifested through ion solvation is therefore of prime interest in chemistry This book considers in depth the phenomenology of ion solvation and the models and interpretations that have been proposed as the physical causes for the observed phenomena It contains a thorough discussion of the statistical thermodynamic background of the solvation process from which a discussion of the actual thermodynamics is developed This in turn serves as a background to the structural and kinetic features of ion solvation

Complex Ions Arthur Jaques, 2015-07-07 Excerpt from Complex Ions In Aqueous Solutions In compiling this volume the needs and criticism of a large class of students unversed in physical chemistry have been especially kept in view and it is considered that the introduction of some elementary matter such as proofs of formulae

which the advanced reader will not require is by no means out of place In giving an account of the methods in Chapters III VI it was found necessary to introduce examples but these were made as brief as possible in order to avoid confusing these chapters with the later ones which deal with practical investigations where more than one method is generally used at a time The tension experiments in Chapter VIII form a method of investigation in which the examination of different salts shows so little variation that it appeared unnecessary to devote a separate chapter to the method The chief aim of the book is to give some account of the more important experimental work in this subject and no apology is offered for the absence of theories of valency Chapter X contains an account of some results besides the identification of complex compounds which have been arrived at by similar methods and which are likely to form the basis of further experiments About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www.forgottenbooks.com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works

Ion Solvation Yizhak Marcus,1985 **Ions , *Ion Channels*** Nikita Gamper,2016-05-01 This Methods in Molecular Biology book offers strategies and protocols for studying a large group of proteins that form ionic channels in the plasma membrane and intracellular membranes of cells Includes step by step protocols materials lists tips and more **Ion the Prize** Leo Sahaya Daphne Antony,2025 Understanding solid liquid interfacial interactions is essential for controlling chemical reactivity material synthesis and electrochemical behaviour These interfaces play a key role in a wide range of applications from energy conversion and storage to the fabrication of functional materials This thesis examines how ions in solution influence the organization of solvent molecules and the energetics at interfaces both in equilibrium and under applied potential The first part of the thesis focuses on the solution based synthesis of low dimensional perovskite materials particularly butylammonium lead iodide It demonstrates that the oxidation state of the lead precursors Pb²⁺ versus Pb⁴⁺ significantly impacts ion solvent complexation leading to distinct crystallization behaviours and morphologies These findings underscore the importance of precursor chemistry in directing material formation at the molecular level The second part of the thesis applies Electrochemical Atomic Force Microscopy to quantitatively map adhesion forces at electrified interfaces with high spatial resolution By examining various electrolyte compositions and electrode materials the thesis reveals how interfacial forces depend on ion identity surface structure and electrochemical conditions These findings provide insights into the nanoscale organization of the electric double layer and uncover ion specific effects that challenge traditional models of electrochemical interfaces This thesis highlights the active and dynamic interplay of all system components including ions solvents and solid surfaces in shaping interfacial structure chemical pathways and electrochemical behaviour **Principles**

of Ionic Organic Reactions Elliot R. Alexander, 2007-03 PREFACE THE Author of this very practical treatise on Scotch Loch Fishing desires clearly that it may be of use to all who had it He does not pretend to have written anything new but to have attempted to put what he has to say in as readable a form as possible Everything in the way of the history and habits of fish has been studiously avoided and technicalities have been used as sparingly as possible The writing of this book has afforded him pleasure in his leisure moments and that pleasure would be much increased if he knew that the perusal of it would create any bond of sympathy between himself and the angling community in general This section is interleaved with blank sheets for the readers notes The Author need hardly say that any suggestions addressed to the case of the publishers will meet with consideration in a future edition We do not pretend to write or enlarge upon a new subject Much has been said and written and well said and written too on the art of fishing but loch fishing has been rather looked upon as a second rate performance and to dispel this idea is one of the objects for which this present treatise has been written Far be it from us to say anything against fishing lawfully practised in any form but many pent up in our large towns will bear us out when we say that on the whole a days loch fishing is the most convenient One great matter is that the loch fisher is dependent on nothing but enough wind to curl the water and on a large loch it is very seldom that a dead calm prevails all day and can make his arrangements for a day weeks beforehand whereas the stream fisher is dependent for a good take on the state of the water and however pleasant and easy it may be for one living near the banks of a good trout stream or river it is quite another matter to arrange for a days river fishing if one is looking forward to a holiday at a date some weeks ahead Providence may favour the expectant angler with a good day and the water in order but experience has taught most of us that the good days are in the minority and that as is the case with our rapid running streams such as many of our northern streams are the water is either too large or too small unless as previously remarked you live near at hand and can catch it at its best A common belief in regard to loch fishing is that the tyro and the experienced angler have nearly the same chance in fishing the one from the stern and the other from the bow of the same boat Of all the absurd beliefs as to loch fishing this is one of the most absurd Try it Give the tyro either end of the boat he likes give him a cast of ally flies he may fancy or even a cast similar to those which a crack may be using and if he catches one for every three the other has he may consider himself very lucky Of course there are lochs where the fish are not abundant and a beginner may come across as many as an older fisher but we speak of lochs where there are fish to be caught and where each has a fair chance Again it is said that the boatman has as much to do with catching trout in a loch as the angler Well we dont deny that In an untried loch it is necessary to have the guidance of a good boatman but the same argument holds good as to stream fishing **Solvation of Ions** Anna King, 1967 *Ions, Electrodes, and Membranes* Jiri Koryta, 1988 **The Ion Channel** E.C. Conley, 1996

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