

1. Read the given statements and select the correct option.

**Statement 1 :** The formula connecting  $u, v$  and  $f$  for a spherical mirror is valid only for mirrors whose sizes are very small compared to their radii of curvature.

**Statement 2 :** Laws of reflection are strictly valid for plane surfaces, but not for large spherical surfaces.

- (A) Both statements 1 and 2 are true and statement 2 is the correct explanation of statement 1.  
(B) Both statements 1 and 2 are true but statement 2 is not the correct explanation of statement 1.  
(C) Statement 1 is true but statement 2 is false.  
(D) Statement 1 is false but statement 2 is true.

2. A boy throws a table tennis ball of mass 20 g upwards with a velocity of  $u_0 = 10$  m/s at an angle  $\theta_0$  with the vertical. The wind imparts a horizontal force of 0.08 N, so that the ball returns to the starting point. Then, the angle  $\theta_0$  must be such that,  $\tan \theta_0$  is  
(A) 0.2      (B) 0.4      (C) 2.5      (D) 1.2

3. A weight is attached to the free end of a sonometer wire. It gives resonance at a length 40 cm when it is resonanced with a tuning fork of frequency 51 Hz. The weight is then immersed


wholly in water, the resonant length is reduced to 30 cm. The relative density in which weight suspended is

- (A) 16/9      (B) 16/7      (C) 16/5      (D) 16/3

4. Hydrogen sulphide ( $\text{H}_2\text{S}$ ) contains 94.11% sulphur, water ( $\text{H}_2\text{O}$ ) contains 11.11% hydrogen and sulphur dioxide ( $\text{SO}_2$ ) contains 50% oxygen. Find the ratio of all given elements. After your calculations which law has been verified?

- (A) Law of multiple proportions  
(B) Law of reciprocal proportions  
(C) Law of constant composition  
(D) Law of combining volumes

5. Which of the following represents the given mode of hybridisation,  $sp^2-sp^2-sp-sp$  from left to right?

- (A)  $\text{CH}_2=\text{CH}-\text{C}\equiv\text{N}$       (B)  $\text{CH}\equiv\text{C}-\text{C}\equiv\text{CH}$   
(C)  $\text{CH}_2=\text{C}=\text{C}=\text{CH}_2$       (D) 

6. In the reaction,



when 1 mole of ammonia and 1 mole of  $\text{O}_2$  are made to react to completion

- (A) 1.0 mole of  $\text{H}_2\text{O}$  is produced  
(B) 2.0 moles of NO will be produced  
(C) All the oxygen will be consumed  
(D) All the ammonia will be consumed.

# National Science Olympiad Sample Papers

**JL Elias**



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## **Table of Contents National Science Olympiad Sample Papers**

1. Understanding the eBook National Science Olympiad Sample Papers
  - The Rise of Digital Reading National Science Olympiad Sample Papers
  - Advantages of eBooks Over Traditional Books
2. Identifying National Science Olympiad Sample Papers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in a National Science Olympiad Sample Papers
  - User-Friendly Interface
4. Exploring eBook Recommendations from National Science Olympiad Sample Papers
  - Personalized Recommendations
  - National Science Olympiad Sample Papers User Reviews and Ratings

- National Science Olympiad Sample Papers and Bestseller Lists
- 5. Accessing National Science Olympiad Sample Papers Free and Paid eBooks
  - National Science Olympiad Sample Papers Public Domain eBooks
  - National Science Olympiad Sample Papers eBook Subscription Services
  - National Science Olympiad Sample Papers Budget-Friendly Options
- 6. Navigating National Science Olympiad Sample Papers eBook Formats
  - ePub, PDF, MOBI, and More
  - National Science Olympiad Sample Papers Compatibility with Devices
  - National Science Olympiad Sample Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of National Science Olympiad Sample Papers
  - Highlighting and Note-Taking National Science Olympiad Sample Papers
  - Interactive Elements National Science Olympiad Sample Papers
- 8. Staying Engaged with National Science Olympiad Sample Papers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers National Science Olympiad Sample Papers
- 9. Balancing eBooks and Physical Books National Science Olympiad Sample Papers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection National Science Olympiad Sample Papers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine National Science Olympiad Sample Papers
  - Setting Reading Goals National Science Olympiad Sample Papers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of National Science Olympiad Sample Papers
  - Fact-Checking eBook Content of National Science Olympiad Sample Papers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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