

Gabriel synthesis



Nitrogen and Phosphorus Containing Compounds

- Stricker synthesis \rightarrow amide + nitrile (amine group + nitrile group) \rightarrow amine acid
- Gabriel synthesis: SN2, hydrolysis, decarboxylation \rightarrow amine acid
- Hydrolytic release of P₄ \rightarrow provides energy
- Phosphoric acid: 3 H⁺, each with unique pKa values \rightarrow buffer over a large pH range

Spectroscopy

- IR - absorbs light of specific wavenumber \rightarrow vibrates (symmetric bonds and stretches)
 - O-H: 3300 broad, N-H: 3300 sharp, C=O: 1700 sharp
- UV - measure absorption of UV light \rightarrow causes movement of electrons in MO
 - more conjugation \rightarrow lower energy for transmission and longer wavelength
 - Smaller gap for HOMO and LUMO \rightarrow move easily from one orbital to another (easily excited) and absorb longer wavelengths (lower frequency, lower energy)
 - More conjugation - absorbs light in visible range
- NMR - excite from alpha (lower energy) state to beta (higher energy) state
 - Left: downfield, deshielded

Separations and Purifications

- Extraction: organic and aqueous phase as separatory funnel \rightarrow mixing
- Filtration: residue, filtrate
- Recrystallisation: heat, cool - recrystallise out of solution
- Distillation: simple: difference $> 25^\circ\text{C}$; superheating when liquid is heated above BP with vaporization, happens when gas bubbles can not overcome column of atmosphere

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Kaplan Test Prep



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