

# CHAPTER 14 STUDY GUIDE

## Mixtures and Solutions

### Section 14.1 Heterogeneous and Homogeneous Mixtures

*In your textbook, read about suspensions and colloids.*

For each statement below, write *true* or *false*.

- \_\_\_\_\_ 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed.
- \_\_\_\_\_ 2. The most abundant substance in a colloid is the dispersion medium.
- \_\_\_\_\_ 3. A colloid can be separated by filtration.
- \_\_\_\_\_ 4. A solid emulsion consists of a liquid dispersed in a solid.
- \_\_\_\_\_ 5. Whipped cream is an example of a foam.
- \_\_\_\_\_ 6. In an aerosol, the dispersing medium is a liquid.
- \_\_\_\_\_ 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
- \_\_\_\_\_ 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
- \_\_\_\_\_ 9. Stirring an electrolyte into a colloid stabilizes the colloid.
- \_\_\_\_\_ 10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter			
12. Between 1 nm and 1000 nm in diameter			
13. More than 1000 nm in diameter			
14. Settle out if undisturbed			
15. Pass through standard filter paper			
16. Lower vapor pressure			
17. Scatter light			

# Mixtures And Solutions Study Guide Answer Key

**Puja Mehta**



## **Mixtures And Solutions Study Guide Answer Key:**

*Study Guide and Full Solutions Manual* Susan McMurry, John McMurry, 2002-08 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all end of chapter problems in the textbook

**Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry** Katherine J. Denniston, Denniston Katherine, 2006      **Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

**Student Study Guide & Selected Solutions Manual** David D. Reid, 2007      **Chemical Interactions**, 2005      Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry John McMurry, Susan McMurry, 1986

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**the Universe** ,1993      *Matter* ,1994 1995 2000 State Textbook Adoption Rowan Salisbury      *Study Guide to Accompany Raven and Johnson Biology* David A. Stetler,1986      eBook: General, Organic and Biological Chemistry 2e  
SMITH,2012-02-16 eBook General Organic and Biological Chemistry 2e      **Prentice Hall Science Explorer** Michael J. Padilla,2002      **Fundamentals of Thermodynamics** Mr. Rohit Manglik,2023-07-23 Explains thermodynamic principles laws and their applications in engineering systems      **Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed** ,2000

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