

Chemistry – Unit 7 Worksheet 3

Write balanced chemical equations for the following reactions.

1. Ammonia (NH_3) reacts with hydrogen chloride to form ammonium chloride.
2. Calcium carbonate decomposes upon heating to form calcium oxide and carbon dioxide.
3. Barium oxide reacts with water to form barium hydroxide.
4. Acetaldehyde (CH_3CHO) decomposes to form methane (CH_4) and carbon monoxide.
5. Zinc reacts with copper(II) nitrate to form zinc nitrate and copper.
6. Calcium sulfite decomposes when heated to form calcium oxide and sulfur dioxide.
7. Iron reacts with sulfuric acid (H_2SO_4) to form iron(II) sulfate and hydrogen gas.
8. A nitrogen containing carbon compound, $\text{C}_2\text{H}_6\text{N}_2$, decomposes to form ethane, C_2H_6 , and nitrogen gas.
9. Phosgene, COCl_2 , is formed when carbon monoxide reacts with chlorine gas.

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