

## Chemistry – Unit 7 Review

### Chemical Reaction Model

1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example).

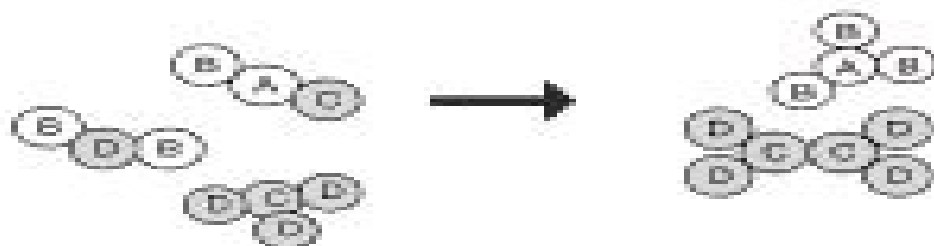
In chemical reactions, atoms of the reactants recombine to form new substances in the products. There is almost always some transfer between the chemical and chemical energy accounts as well as an exchange of thermal energy via heating/cooling between the system and surroundings. The balanced chemical equation represents this process symbolically, showing that the atoms in the products are the same as those in the reactants.



Note: there are 4 atoms of H and 2 atoms of O on each side of the equation.

### Applying the Model

2. Balance the reaction below by drawing any additional reactant and/or product molecules that would be needed.



Write the balanced equation for this reaction.



Which of these molecules would be called reactants?

$2\text{B}_2$ ,  $\text{C}_2\text{B}_2$ , and  $\text{BAC}$  are reactants.

# Modeling Chemistry Unit 7 Test Answers

**Emilie Sanchez**



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