

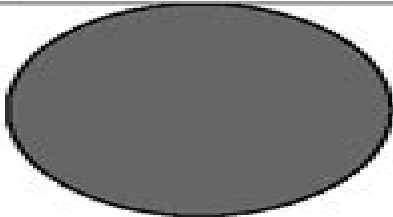


Atomic Theory Worksheet

Objective: You will create an abbreviated atomic theory timeline to show how the scientists involved in the atomic theory worked together to construct the modern model of the atom.

Part 1 – Atomic Theory Timeline

Choose 3 scientists from the options below and complete the table.

Options: John Dalton, Niels Bohr, JJ Thomson, Erwin Schrödinger, Ernest Rutherford

Content	Scientist 1	Scientist 2	Scientist 3
Timeframe [in order]	1803	1897	1911
Scientist Name	John Dalton	J.J. Thompson	Ernest Rutherford
What did he contribute to the atomic theory?	He contributed that atoms can not be made or destroyed, a chemical reaction only can happen when atoms are rearranged, as well all atoms of a specific element are the same in mass and properties.	He contributed that atoms have charged particles that are called electrons, he also thought that matter is electronically neutral, he also thought that electrons are dispersed evenly through an atom.	He contributed that an atom has a very small positively charged center called the nucleus and it contains most of the atoms mass. He also added that atoms have a cloud of electrons that circle around the nucleus.
Draw a sketch of their model [by hand or using the drawing tool] and label all parts.		 atom electrons	 electrons nucleus

Part 2 – Reflection Questions

1. Describe two details of the modern atomic model.

Atoms have a nucleus of protons and neutrons and a cloud around the nucleus that contains the electrons.

2. Explain how the atomic model has changed over time.

The Rutherford model got replaced by the Bohr model because scientists found out that electrons have a specific orbit around the nucleus.

Modern Atomic Theory Concept Review Answers

Jianjun Gao



Modern Atomic Theory Concept Review Answers:

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