- Fill in the gaps used in the following text by adding suffix prefix or both with the root words underlined in the text:

 0.5 × 8 = 4

 We are highly (a) grate to our courageous freedom (b) fight For their (c) hero deeds in 1971, we achieved our (d) free. But this is so (e) fortunate that many of them are now leading an (f) human life. Our government should be (g) care about developing their status. It's really (h) shame for us that even after 42 years of independence, they are leading a miserable life.
- 8. Complete the following text with suitable verb in the box with right form.

							$0.0 \times 0 = 4$
Obtain	Go	Help	Deny	Regard	Take	Redefin	depend
Education	is (a)	as th	ne yardsti	ck of dev	elopmer	it. Develop	oment and
education	(b)	hand in	hand. (c)	degre	e is not	the prime	object of
education.	Educat	ion has to	o be (d) _	_ It is high	n time w	re (e) m	easures to
spread ed	ucation	. Only t	he educa	ted parent	ts (f) _	_ their c	hildren in
education.	There	fore, non	ne (g)	the fact	that the	foundation	on of any
country is	largely	(h) u	pon the ed	lucation of	its yout	h.	504 Metarina (10 milio 4 1.)

 $0.5 \times 8 = 4$

Section B: Composition, Marks-20

- 9. Suppose you are Nabil/Nabila and you are a student of 'x' Zilla school, Khulna. There is no common room in your school. For want of a common room the students waste their leisure hours aimlessly. Now write an application to your headmaster on behalf of the students requesting him to set up a common room in your school.
- 10. Write a composition on the following topic in 250 words: 12
 "Your Future Plan of life"

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C Cleary

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web a which chemical is the limiting reactant b how many grams of zns will be formed c how many grams of the excess reactant will remain after the reaction is over 3 which element is in excess when 3 00 grams of mg is ignited in 2 20 grams of pure oxygen what mass is in excess what mass of mgo is formed 4

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web the limiting reactant or limiting reagent is the reactant that gets consumed first in a chemical reaction and therefore limits how much product can be formed as we saw in example 1 there are many different ways to determine the limiting reactant but they all involve using mole ratios from the balanced chemical equation

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