

## Chemistry – Unit 6 Worksheet 1

After having observed molecular models which show 18.098 compounds, you discovered that certain particles are known to form positively charged ions, but cannot, while some particles are known to form negatively charged ions (anions). However, many atoms cannot make continuous particles, and a substance has been experimentally shown that such molecules do not form. After all, these examples are positive that some of the atoms in which these elements combine to form the compounds that change off their own shape.

1. After that formula, what does the particle structure that shows composition. The same and ions are shown in the table.

18.098 molecular formula	1. molecule 2. oxygen	2. molecule 3. oxygen	2. molecule 3. oxygen	2. molecule 3. oxygen
	$\text{H}_2\text{O}$	$\text{H}_2\text{O}$	$\text{H}_2\text{O}$	$\text{H}_2\text{O}$

particle  
structure



18.098 molecular formula	2. molecule 3. oxygen	2. molecule 3. oxygen	2. molecule 3. oxygen
	$\text{H}_2\text{O}$	$\text{H}_2\text{O}$	$\text{H}_2\text{O}$

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18.098 molecular formula	2. molecule 3. oxygen	2. molecule 3. oxygen	2. molecule 3. oxygen	2. molecule 3. oxygen
	$\text{H}_2\text{O}$	$\text{H}_2\text{O}$	$\text{H}_2\text{O}$	$\text{H}_2\text{O}$

particle  
structure



# Modeling Chemistry U6 Ws1 V2 0

**N Colangelo**



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