

Chemistry – Unit 7 Worksheet 3

Write balanced chemical equations for the following reactions.

- a. Aluminum metal reacts with hydrochloric acid to form aluminum chloride.
- b. Acetylene (C_2H_2), a hydrocarbon, burns with oxygen (O_2) and carbon dioxide.
- c. Zinc reacts with copper(II) sulfate to form zinc sulfate and copper.
- d. Ethanol reacts with oxygen to form carbon dioxide and water.
- e. Sodium metal reacts with water to form sodium hydroxide.
- f. Nitrogen (N_2) is placed in a hot glowing crucible until it is the gas.
- g. Magnesium metal burns when exposed to light to form magnesium hydroxide.
- h. Nitrogen gas reacts with water to produce ammonia (NH_3).

Modeling Chemistry Unit 7 Ws 4 V2

TD Snyder



Modeling Chemistry Unit 7 Ws 4 V2:

Modeling Chemistry Unit 7 Ws 4 V2 Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Modeling Chemistry Unit 7 Ws 4 V2**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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