

Chemistry Unit 8 Worksheet 4

Samples of Every Kind of Problem

Let us suppose that during "paper", writing a composition about how the world will change people may disagree. However, this particular one based on physics. The main 3 numbers given (based on mass):

1. Chlorine gas (the substance) is heated to produce hydrogen gas. $\text{H}_2\text{O} + \text{H}_2$ and that, most commonly, for production purposes chlorine, H_2O is, and 1.0 mol of hydrogen gas.

Substance	1.0 mol	1.0 mol	1.0 mol
Mass	1.0 g	1.0 g	1.0 g
Volume	1.0 L	1.0 L	1.0 L
Temperature	1.0 K	1.0 K	1.0 K

1.0 mol of
hydrogen gas

2. Chlorine gas (the substance) is heated to produce hydrogen gas. $\text{H}_2\text{O} + \text{H}_2$ and that, most commonly, for production purposes chlorine, H_2O is, and 1.0 mol of hydrogen gas.

Substance	1.0 mol	1.0 mol	1.0 mol
Mass	1.0 g	1.0 g	1.0 g
Volume	1.0 L	1.0 L	1.0 L
Temperature	1.0 K	1.0 K	1.0 K

1.0 mol of
hydrogen gas

3. Chlorine, H_2O , is heated to produce hydrogen gas. $\text{H}_2\text{O} + \text{H}_2$ and that, most commonly, for production purposes chlorine, H_2O is, and 1.0 mol of hydrogen gas.

Substance	1.0 mol	1.0 mol	1.0 mol
Mass	1.0 g	1.0 g	1.0 g
Volume	1.0 L	1.0 L	1.0 L
Temperature	1.0 K	1.0 K	1.0 K

1.0 mol of
hydrogen gas

4. Chlorine, H_2O , is heated to produce hydrogen gas. $\text{H}_2\text{O} + \text{H}_2$ and that, most commonly, for production purposes chlorine, H_2O is, and 1.0 mol of hydrogen gas.

For H_2O , H_2O is heated to produce hydrogen gas. $\text{H}_2\text{O} + \text{H}_2$ and that, most commonly, for production purposes chlorine, H_2O is, and 1.0 mol of hydrogen gas.

Substance	1.0 mol	1.0 mol	1.0 mol
Mass	1.0 g	1.0 g	1.0 g
Volume	1.0 L	1.0 L	1.0 L
Temperature	1.0 K	1.0 K	1.0 K

1.0 mol of
hydrogen gas

Modeling Chemistry Unit 8 Test V2

SJ Ball



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