

# S.L.O.P - Paper 1

Periodic table on back!

## Unit 3 Quantitative Chemistry

Name:

Class:

### Number of elements in a formula

**Example:**  $\text{FeBr}_3$

The number of elements is the same as the number of symbols.  
Each different symbol starts with a capital letter.



So we have **two** different elements **Fe** = Iron, **Br** = Bromine

**Task 1:** How many different **elements** are in each chemical formula?



### Number of atoms of each element in a formula

**Example:**



The small numbers on the bottom right of each symbol tell you how many atoms of that element you have. If there is no number to the right you have one atom of that element.



**Task 2:** How many atoms of each element are in these formulae?  
Also give the total number of atoms:



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**Robert R. Redfield**



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## **Table of Contents L O Paper Exam Gradenovember2013**

1. Understanding the eBook L O Paper Exam Gradenovember2013
  - The Rise of Digital Reading L O Paper Exam Gradenovember2013
  - Advantages of eBooks Over Traditional Books
2. Identifying L O Paper Exam Gradenovember2013
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an L O Paper Exam Gradenovember2013
  - User-Friendly Interface
4. Exploring eBook Recommendations from L O Paper Exam Gradenovember2013

- Personalized Recommendations
  - L O Paper Exam Gradenovember2013 User Reviews and Ratings
  - L O Paper Exam Gradenovember2013 and Bestseller Lists
5. Accessing L O Paper Exam Gradenovember2013 Free and Paid eBooks
- L O Paper Exam Gradenovember2013 Public Domain eBooks
  - L O Paper Exam Gradenovember2013 eBook Subscription Services
  - L O Paper Exam Gradenovember2013 Budget-Friendly Options
6. Navigating L O Paper Exam Gradenovember2013 eBook Formats
- ePub, PDF, MOBI, and More
  - L O Paper Exam Gradenovember2013 Compatibility with Devices
  - L O Paper Exam Gradenovember2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of L O Paper Exam Gradenovember2013
  - Highlighting and Note-Taking L O Paper Exam Gradenovember2013
  - Interactive Elements L O Paper Exam Gradenovember2013
8. Staying Engaged with L O Paper Exam Gradenovember2013
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers L O Paper Exam Gradenovember2013
9. Balancing eBooks and Physical Books L O Paper Exam Gradenovember2013
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection L O Paper Exam Gradenovember2013
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine L O Paper Exam Gradenovember2013
- Setting Reading Goals L O Paper Exam Gradenovember2013
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of L O Paper Exam Gradenovember2013

- Fact-Checking eBook Content of L O Paper Exam Gradenovember2013
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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