# III BOILE Ionic Bonding

Metals and non-metals react together to form \_\_\_\_\_ compounds. They are held together by ionic bonds. Ionic bonding is a force of \_\_\_\_\_\_ attraction between two \_\_\_\_ atoms (called \_\_\_\_). An atom becomes charged by either donating or accepting \_\_\_ electrons from another atom. A valence electron is one from the \_\_\_\_\_\_ shell of the atom.

Since opposites attract, there is an attraction between the positively charged and negatively charged ions that holds them together.

If an atom \_\_\_\_ an electron, it becomes positively charged. This is then called a \_\_\_\_ always donate electrons. If an atom accepts an electron, it becomes charged. This is called an \_\_\_\_\_\_ Non-metals always accept electrons.

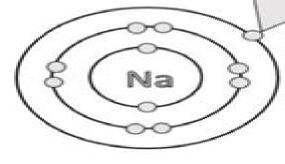
Atoms do this in order to become '\_\_\_\_\_. A stable atom is one \_\_\_\_\_ of electrons

which has obtained a \_\_\_\_\_\_ of electrons.

How do we know which atoms donate and which accept electrons?

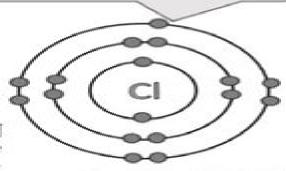
Sodium has \_\_\_\_ lone electron in the outer shell. It is easiest for this atom to \_\_\_\_\_\_ one electron than to try and find seven more to fill the outer shell

Chlorine has \_\_\_\_\_ electrons in its outer shell. Therefore, it is easiest for Chlorine to try and donate seven.



donate electrons and therefore become cations.

> accept electrons and therefore become anions.



# **Ionic Bond Worksheet Teaching Transparency**

**Thomas Griffiths** 

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Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004 Glencoe Science McGraw-Hill Staff,2001-08 Te HS&T | Holt Rinehart & Winston, Holt, Rinehart and Winston **Chemistry** John S. Phillips, Cheryl Wistrom, 2000 Staff, 2004-02 Prentice Hall Science Explorer Michael J. Padilla, 2002 The Science Teacher, 2009 The School Science Review .2000 **Earth Science** Samuel N. Namowitz, Nancy E. Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Science Interactions** Spaulding, 1989 Robert W. Avakian, 1995-07-17 Carolina Science and Math Carolina Biological Supply Company, 2003 Glencoe Earth Science Ralph M. Feather, 1999 Earth science is the study of Earth and space It is the study of such things as the transfer of energy in Earth's atmosphere the evolution of landforms patterns of change that cause weather the scale and structure of stars and the interactions that occur among the water atmosphere and land Earth science in this book is divided into four specific areas of study geology meteorology astronomy and oceanography p 8 9 School Library Journal ,1988 Teaching Chemical Bonding Margaret Irene Lindsay, 1995 This document presents an instructional strategy for teaching chemical bonding using parables and music Games student interactions and worksheets are included in the lesson plans Topics include metallic bonding covalent bonding including molecular and network structure and ionic bonding IRH

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